

KISHORE VAIGYANIK PROTSAHAN YOJANA - 2013

Date: 27-10-2013 Duration: 3 Hours Max. Marks: 100

STREAM - SA

GENERAL INSTRUCTIONS

- The Test Booklet consists of 80 questions.
- There are Two parts in the question paper. The distribution of marks subjectwise in each part is as under for each correct response.

MARKING SCHEME:

PART-I:

MATHEMATICS

Question No. 1 to 15 consist of ONE (1) mark for each correct response.

PHYSICS

Question No. 16 to 30 consist of ONE (1) mark for each correct response.

CHEMISTRY

Question No. 31 to 45 consist of ONE (1) mark for each correct response.

BIOLOGY

Question No. 46 to 60 consist of ONE (1) mark for each correct response.

PART-II:

MATHEMATICS

Question No. 61 to 65 consist of TWO (2) marks for each correct response.

PHYSICS

Question No. 66 to 70 consist of TWO (2) marks for each correct response.

CHEMISTRY

Question No. 71 to 75 consist of TWO (2) marks for each correct response.

BIOLOGY

Question No. 76 to 80 consist of TWO (2) marks for each correct response.

PART-I One Mark Questions

MATHEMATICS

- 1. Let x, y, z be three non-negative integers such that x + y + z = 10. The maximum possible value of xyz + xy + yz + zx is
 - (A) 52

- (B)64
- (C) 69

(D) 73

Sol. Taking three no's.

$$x + 1$$
, $y + 1$, $z + 1$

 $AM \ge GM$.

$$\frac{\left(x+1\right)+\left(y+1\right)+\left(z+1\right)}{3} \geq \left\{\!\left(x+1\right)\!\!\left(y+1\right)\!\!\left(z+1\right)\!\!\right\}^{1/3}$$

$$\left(\frac{13}{3}\right)^3 \ge xyz + xy + yz + zx + 11$$

$$\left(\frac{13}{3}\right)^3 - 11 \ge xyz + xy + yz + zx$$

equality hold when x = y = z but x + y + z = 0 and x, y, z are integers.

So maximum value when any two of x, y, z are equal to 3 and third is equal to 4.

Ans. (C)

- 2. If a, b are natural numbers such that $2013 + a^2 = b^2$, then the minimum possible value of ab is (A) 671 (B) 668 (C) 658 (D) 645
 - $(b-a)(b+a) = 2013 = 3 \times 11 \times 61$

ab minimum when b - a = 33

$$a = 14$$

$$ab = 14 \times 47 = 658$$

Ans. (C)

- The number of values of b for which there is an isosceles triangle with sides of length b + 5, 3b 2 and 6 b is (A) 0 (B) 1 (C) 2 (D) 3
- Sol. Case (I)

Sol.

$$b + 5 = 3b - 2$$

$$\Rightarrow$$
 b = $\frac{7}{2}$

So sides are
$$\frac{17}{2}, \frac{17}{2}, \frac{5}{2}$$

Case (II)

$$b + 5 = 6 - b = b = \frac{1}{2}$$

Sides
$$\frac{11}{2}$$
, $\frac{-1}{2}$, $\frac{11}{2}$ Not possible

Case (III)
$$3b - 2 = 6 - b$$

$$4b = 8$$

$$b = 2$$

two cases are possible

4. Let a, b be non-zero real numbers. Which of the following statements about the quadratic equation $ax^2 + (a + b)x + b = 0$

is neccesarily true?

- (I) It has at least one negative root
- (II) It has at least one positive root.
- (III) Both its roots are real.
- (A) (I) and (II) only
- (B) (I) and (III) only
- (C) (II) and (III) only
- (D) All of them

Sol. $ax^2 + (a + b)x + b = 0$

$$(x + 1) (ax + b) = 0 \text{ roots are } -1, \frac{-b}{a}$$

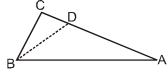
Ans. (B)

Let x, y, z be non-zero real numbers such that $\frac{x}{y} + \frac{y}{z} + \frac{z}{x} = 7$ and $\frac{y}{x} + \frac{z}{y} + \frac{x}{z} = 9$, then $\frac{x^3}{v^3} + \frac{y^3}{z^3} + \frac{z^3}{x^3} - 3$ is 5.

equal to

- (D) 155
- (A) 152 (B) 153 (D) 154 $a^3 + b^3 + c^3 3abc = [a + b + c] [(a + b + c)^2 3(ab + bc + ca)]$ Sol. $= [7][(7)^2 - 3(9)]$ $= 7(49 - 27) = 7 \times 22 = 154$ Ans. (C)
- In a triangle ABC with $\angle A < \angle B < \angle C$, points D, E, F are on the interior of segments BC, CA, AB, respectively. 6. Which of the following triangles CANNOT be similar to ABC?
 - (A) Triangle ABD
- (B) Triangle BCE
- (C) Triangle CAF
- (D) Triangle DEF

Sol.

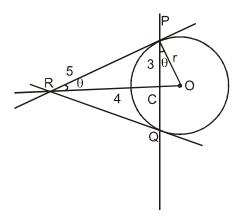


In \triangle ABD greatest angles is \angle D which is greater by \angle C so \triangle ABD is not similar to \triangle ABC.

Ans. (A)

- 7. Tangents to a circle at points P and Q on the circle intersect at a point R. If PQ = 6 and PR = 5 then the radius of the circle is
 - (A) $\frac{13}{3}$
- (B)4

(C) $\frac{15}{4}$



Sol.

In
$$\triangle RCP \Rightarrow \cos\theta = \frac{4}{5}$$

In
$$\triangle PCO \Rightarrow \cos\theta = \frac{3}{r}$$

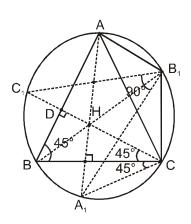
8. In an acute-angled triangle ABC, the altitudes from A, B, C when extended intersect the circumcircle again at points A_1 , B_1 , C_2 , respectively. If $\angle ABC = 45^\circ$ then $\angle A_1B_2$, equals

(A) 45°

(B) 60°

(C) 90°

(D) 135°



Sol.

$$\angle$$
BCH = 45° = \angle BCA₁
 \angle C₁CA₁ = \angle C₁B₁A₁ = 90°
Ans. (C)

9. In a rectangle ABCD, points X and Y are the midpoints of AD and DC, respectively. Lines BX and CD when extended intersect at E, lines BY and AD when extended intersect at F. If the area of ABCD is 60 then the area of BEF is

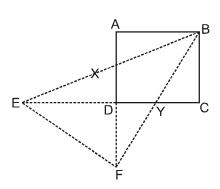
(A)60

(B) 80

(C) 90

(D) 120

Sol.



Let AB = x

$$BC = y$$

$$xy = 60$$

∆XED ~∆XBA

and $\Delta DFY \simeq \Delta CBY$

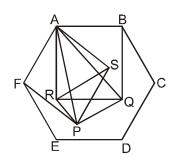
$$\Delta BEF = \mathbf{I}BXDY + \Delta XED + \Delta DFY + \Delta DEF$$

= $60 + \Delta DEF$

$$60 + \frac{1}{2} \times xy$$

60 + 30 = 90

10. In the figure given below, ABCDEF is a regular hexagon of side length 1, AFPS and ABQR are squares. Then the ratio Area (APQ)/ Area (SRP) equals



(A)
$$\frac{\sqrt{2}+1}{2}$$

(B) $\sqrt{2}$

(C)
$$\frac{3\sqrt{3}}{4}$$

(D) 2



Sol.

$$AP = AQ = \sqrt{2}$$
, $\angle APQ = 30^{\circ}$

In ∆SRP

$$SR = SP = 1$$
, $\angle RSP = 30^{\circ}$

$$\angle$$
BAS = \angle FAB - \angle FAS = 120° - 90° = 30°

$$\angle$$
SAR = \angle BAR - \angle BAS = 60°

In ∆ARS

$$\cos 60^{\circ} = \frac{1+1-SR^2}{2.1.1} [:: AR = AS = 1]$$

$$\Rightarrow$$
 SR = 1

Now,
$$\angle$$
RSP = \angle ASP - \angle ASR

=
$$90^{\circ} - 60^{\circ}$$
 = 30° [: ASR is equilateral]

Now, from
$$\triangle SRP \Rightarrow RP = \frac{\sqrt{3}-1}{2\sqrt{2}}$$

In ∆APR

$$\cos \angle RAP = \frac{(\sqrt{2})^2 + 1^2 - PR^2}{2\sqrt{2}}$$

$$\Rightarrow$$
 cos \angle RAP = 15°

$$\angle PAQ = \angle RAQ - \angle RAP = 45^{\circ} - 15^{\circ} = 30^{\circ}$$

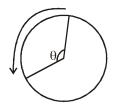
Now,
$$\frac{\operatorname{ar}(\Delta \mathsf{APQ})}{\operatorname{ar}(\Delta \mathsf{SRP})} = \frac{\frac{1}{2} \times \sqrt{2} \times \sqrt{2} \sin 30^{\circ}}{\frac{1}{2} \times 1 \times 1 \times \sin 30^{\circ}} = 2$$

Ans. (D)

- 11. A person X is running around a circular track completing one round every 40 seconds. Another person Y running in the opposite direction meets X every 15 second. The time, expressed in seconds, taken by Y to complete one round is
 - (A) 12.5
- (B) 24
- (C) 25

(D) 55

Sol.



$$\theta = \frac{2\pi}{40} \times 15 = 2\pi - \frac{2\pi}{n} \times 15$$

$$\therefore \frac{3}{8} = 1 - \frac{15}{n}$$

Ans. (B)

12. The least positive integer n for which

$$\sqrt{n+1} - \sqrt{n-1} < 0.2$$
 is

(A) 24

- (B) 25
- (C) 26

(D) 27

Sol. $\sqrt{n+1} < 0.2 + \sqrt{n-1}$

squaring

$$1.96 < .4 \sqrt{n-1}$$

again squaring $(4.9)^2 + 1 < n$

Ans. (C)

- **13.** How many natural numbers n are there such that n!+10 is a perfect square?
 - (A) 1

(B) 2

(C)4

(D) infinitely many

Sol. If n = 1, 2, 4, 5, n! + 10 is not a perfect square

If n = 3, n! + 10 is a perfact square

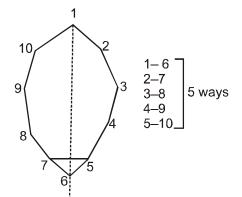
If n > 5

n! + 10 =
$$\frac{10}{1}$$
 [3 × 4 × 6xn + 1]
is odd

than exponant of 2 is one so it is not a perfact square.

Ans. (A)

- 14. Ten points lie in a plane so that no three of them are collinear. The number of lines passing through exactly two of these points and dividing the plane into two regions each containing four of the remaining points is
 - (A) 1
 - (B) 5 (C) 10
 - (D) dependent on the configuration of points



Sol.

Ans. (B)

- 15. In a city, the total income of all people with salary below Rs. 10000 per annum is less than the total income of all people with salary above Rs. 10000 per annum. If the salaries of people in the first group increases by 5% and the salaries of people in the second group decreases by 5% then the average income of all people (A) increases
 - (C) remains the same (D) cannot be determined from the data
- Sol. Let total number of people with salary below Rs. 10000 per annum is x and salary is A. Let total number of people with salary above Rs. 10000 per annum is y and salary is B then.

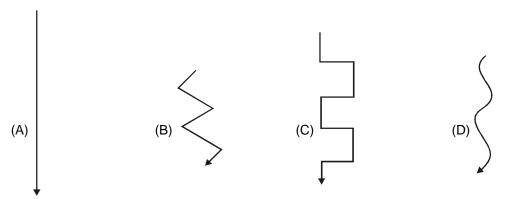
$$\frac{\text{average after}}{\text{average before}} = \frac{x \left(\frac{105}{100}A\right) + y \left(\frac{95}{100}B\right)}{xA + yB}$$

$$= 1 + \frac{5}{100} \left(\frac{xA - yB}{xA + yB} \right)$$

Ans. (B)

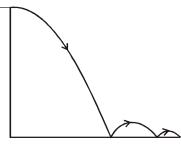
PHYSICS

16. A man inside a freely falling box throws a heavy ball towards a side wall. The ball keeps on bouncing between the opposite walls of the box. We neglect air resistance and friciton. Which of the following figures depicts the motion of the centre of mass of the entire system (man, the ball and the box)?

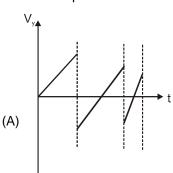


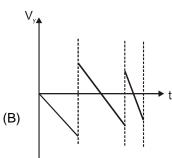
Sol. CM will go downwards Ans. (A)

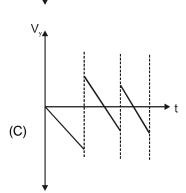
17. A ball is thrown horizontally from a height with a certain initial velocity at time t = 0. The ball bounces repeatedly from the ground with the coefficient of restitution less than 1 as shown.

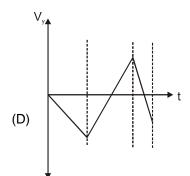


Neglect air resistance and taking the upward direction as positive, which figure qualitatively depicts the vertical component of the ball's velocity (V_x) as a function of time (t)?

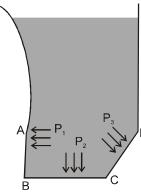








- as $V_y = U_y gt$ Ans. (B) Sol.
- 18. A tall tank filled with water has an irregular shape as shown. The wall CD makes an angle of 45° with the horizontal; the wall AB is normal to the base BC. The lengths AB and CD are much smaller than the height h of water (figure not to scale).



Let P_1 , P_2 and P_3 be the pressures exerted by the water on the wall AB, base BC and the wall CD respectively. Density of water is ρ and g is acceleration due to gravity. Then, approximately

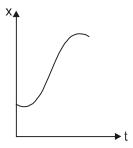
(A)
$$P_1 = P_2 = P_3$$

(A)
$$P_1 = P_2 = P_3$$
 (B) $P_1 = 0$, $P_3 = \frac{1}{\sqrt{2}}P_2$ (C) $P_1 = P_3 = \frac{1}{\sqrt{2}}P_2$ (D) $P_1 = P_3 = 0$, $P_2 = h\rho g$

(D)
$$P_1 = P_3 = 0$$
, $P_2 = h\rho g$

- Sol. All are nearly at same height hence $P_1 = P_2 = P_3$
 - Ans. (A)

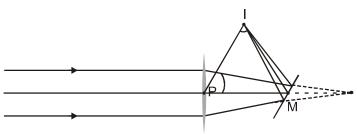
19. The accompanying graph of position x versus time t represents the motion of a particle. If p and q are both positive constants, the expression that best describes the acceleration α of the particle is



- (A) a = -p qt
- (B) a = -p + qt
- (C) a = p + qt
- (D) a = p qt

- From graph V first increases then decreases Sol. Hence a is earliar positive then negative
 - a = P qt
 - Ans. (D)
- 20. Two stones of mass m_1 and m_2 (such that $m_1 > m_2$) are dropped Δt time apart from the same height towards the ground. At a later time t the difference in their speed is ΔV and their mutual separation is ΔS . While both stones are in flight
 - (A) ΔV decreases with time and ΔS increases with time
 - (B) Both ΔV and ΔS increase with time
 - (C) ΔV remains constant with time and ΔS decreases with time
 - (D) ΔV remains constant with time and ΔS increases with time
- Sol. ΔV = Const & ΔS increases with time
 - Ans. (D)
- The refractive index of a prism measured using three lines of a mercury vapour lamp. If μ_1 , μ_2 and μ_3 are the measured refractive indices for these green, blue and yellow lines respectively, then 21.
 - (A) $\mu_2 > \mu_3 > \mu_1$
- (B) $\mu_2 > \mu_1 > \mu_3$ (C) $\mu_3 > \mu_2 > \mu_1$ (D) $\mu_1 > \mu_2 > \mu_3$

- $\mu(\lambda) = B + \frac{C}{\lambda^2} + \dots$ Sol.
 - $\mu_2 > \mu_1 > \mu_3$ Ans. (B)
- 22. A horizontal parallel beam of light passes through a vertical convex lens of focal length 20 cm and is then reflected by a tilted plane mirror so that it converges to a point I. The distance PI is 10 cm.



M is a point at which the axis of the lens intersects the mirror. The distance PM is 10 cm. The angle which the mirror makes with the horizontal is

- (A) 15°
- (B) 30°
- $(C) 45^{\circ}$

(D) 60°



Ans. 60°

- 23. In a car a rear view mirror having a radius of curvature 1.50 m forms a virtual image of a bus located 10.0 m from the mirror. The factor by which the mirror magnifies the size of the bus is close to (A) 0.06 (B) 0.07 (C) 0.08 (D) 0.09
- Sol. (A) 0.06u = -10 mR = 1.5 m

$$\frac{1}{y} + \frac{1}{H} = \frac{2}{R}$$

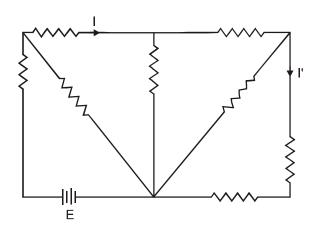
$$\frac{1}{v} - \frac{1}{10} = \frac{2}{1.5}$$

$$v = \frac{30}{43}$$

$$m = -\frac{v}{u} = \frac{30}{43 \times 10} = 0.07$$

Ans. (B)

24. Consider the circuit shown in the figure below:



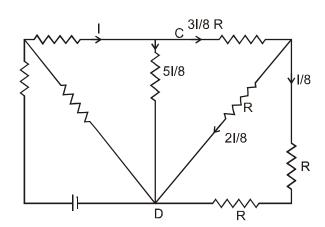
All the resistors are identical. The ratio I/I' is

(A) 8

(B) 6

(C)5

(D) 4



Sol.

$$\frac{1}{l'} = 8$$

Ans.(A)

(D) (I), (II), (III), and (IV)

- **25.** The figure shows a bar magnet and a metallic coil. Consider four situations.
 - (I) Moving the magnet away from the coil.
 - (II) Moving the coil towards the magnet.
 - (III) Rotating the coil about the vertical diameter.
 - (IV) Rotating the coil about its axis.



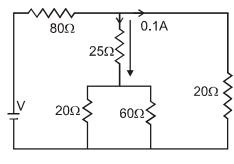
An emf in the coil will be generated for the following situations.

- (A) (I) and (II) only (B) (I), (II) and (IV) only (C) (I), (II), and (III) only
- No EME Induce if ring retate about its own axis (v. 4 ± 0)
- **Sol.** No EMF Induce if ring rotate about its own axis ($:\Delta \phi = 0$)

Hence, I, II & IV are correct

Ans. (C)

26. A current of 0.1 A flows through a 25 Ω resistor represented by the circuit diagram. The current in the 80 Ω resistor is

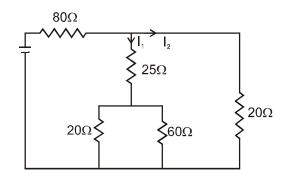


(A) 0.1 A

(B) 0.2 A

(C) 0.3 A

(D) 0.4 A



Sol.

$$0.1 \times \left(25 + \frac{20 \times 60}{20 + 60}\right) = i_2 \times 20$$

 $I_{2} = 0.2A$

Hence, i through 80Ω

0.1 + 0.2 = 0.3A

Ans. (C)

- Solar energy is incident normally on the earth's surface at the rate of about 1.4 kW m⁻². The distance between the earth and the sun is 1.5×10^{11} m. Energy (E) and mass (m) are related by Einstein equation E=mc² where c (3 × 10⁸ ms⁻¹) is the speed of light in free space. The decrease in the mass of the sun is (A) 10^9 kg s⁻¹ (B) 10^{30} kg s⁻¹ (C) 10^{26} kg s⁻¹ (D) 10^{11} kg s⁻¹
- Sol. E Radiated by Sun

 $E = 4\pi r^2 \times 1.4 \text{ kW} = mC^2$

 $E = 4\pi \times (1.5 \times 10^{11})^2 \times 1.4 \times 10^3 = m.(3 \times 10^8)^2$

$$m = \frac{4 \times 22 \times (1.5)^2 \times 1.4 \times 10^9}{7 \times 9} = 10^9 \text{kg/s}$$

Ans. (A)

- 28. If the current through a resistor in a circuit increases by 3%, the power dissipated by the resistor
 - (A) increases approximately by 3%
 - (B) increases approximately by 6%
 - (C) increases approximately by 9%
 - (D) decreases approximately by 3%
- **Sol.** $P = I^2R$

$$\frac{\Delta P}{P} = \frac{2\Delta I}{I} = 6\%$$

Ans. (B)

- 29. An ideal gas filled in a cylinder occupies volume V. The gas is compressed isothermally to the volume V/3. Now the cylinder valve is opened and the gas is allowed to leak keeping temperature same. What percentage of the number of molecules escape to bring the pressure in the cylinder back to its original valuye.
 - (A) 66%
- (B) 33%
- (C) 0.33%
- (D) 0.66%

Sol. $\frac{P_1V_1}{n_1} = \frac{P_2V_2}{n_2}$

$$\frac{P_1V}{n_1}=\frac{P_1.\frac{V}{3}}{n_2}$$

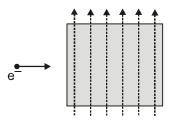
$$\Rightarrow$$
 $n_2 = \frac{n_1}{3}$

Now, $\frac{2}{3}$ of Gas will come out to make the presence P₁

Hence 66.66%

Ans. (A)

30. An electron enters a chamber in which a uniform magnetic field is present as shown



An electric field of appropriate magnitude is also applied so that the electron travels undeviated without any change in its speed thorugh the chamber. We are ignoring gravity. Then, the direction of the electric field is

- (A) opposite to the direction of the magnetic field
- (B) opposite to the direction of the electron's motion
- (C) normal to the plane of the paper and coming out of the plane of the paper
- (D) normal to the plane of the paper and into the plane of the paper

Sol.
$$q \stackrel{\rightarrow}{E} + q \stackrel{\rightarrow}{V} \times \stackrel{\rightarrow}{B} = 0$$

Hence, into the paper

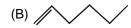
Ans. (D)

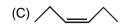
- 31. The moelcule having a formyl group is
 - (A) acetone
- (B) acetaldehyde
- (C) acetic acid
- (D) acetic anhydride

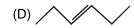
- (a) $CH_3 C CH_3$ \parallel O Sol.

 - (c) $CH_3 C OH$ \parallel O
 - $\begin{array}{c|c} \text{(d) } \operatorname{CH_3} \operatorname{C-O} \operatorname{C-CH_3} \\ \parallel & \parallel \\ \operatorname{O} & \operatorname{O} \end{array}$
 - $\mathrm{CH_3} \mathrm{C-H}$ has formal group.
 - Ans. (B)
- 32. The structure of cis-3-hexene is

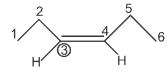








Sol.



Ans. (C)

33. The number of sp² hybridized carbon atoms in

$$\label{eq:hc} \begin{array}{c} O \\ || \\ || \\ C - CH_2 - C - CH_2 - CH = CH_2 \end{array}, \text{ is}$$

(C)4

(D) 6

sp sp sp³
$$\parallel$$
 sp³ sp² sp²
Sol. $HC \equiv C - CH_2 - C - CH_2 - CH = CH_2$

Ans. (A)

- 34. The number of valence electrons in an atom with electronic configuration 1s² 2s² 2p⁶ 3s² 3p³ is
 - (A)2

- (B)3
- (C)5

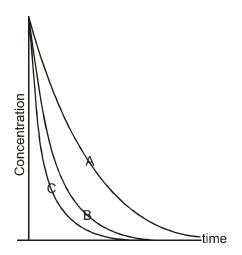
(D) 11

Ans. (C)

- The pair of atoms having the same number of neutrons is 35.
 - (A) ${}^{12}_{6}$ C, ${}^{24}_{12}$ Mg
- (B) $^{23}_{11}$ Na, $^{19}_{9}$ F (C) $^{23}_{11}$ Na, $^{24}_{12}$ Mg
 - (D) ²³₁₁Na,³⁹₁₉K

- Which of the following molecules has no dipole moment? 36.
- (B) CHCl₃
- (C) CH₂Cl₂
- (D) CCI
- Sol. CCI, has zero dipole moment due to its tetrahedral shape, all C-CI bond moment cancel each other. Ans. (D)

37. The decay profiles of three radioactive species A, B and C are given below:



These profiles imply that the decay constants $k_{_A}$, $k_{_B}$ and $k_{_C}$ follow the order (A) $k_{_A} > k_{_B} > k_{_C}$ (B) $k_{_A} > k_{_C} > k_{_B}$ (C) $k_{_B} > k_{_A} > k_{_C}$ (D) $k_{_C} > k_{_B} > k_{_A}$

(A)
$$k_{A} > k_{B} > k_{C}$$

 $C_{t} = C_{0}e^{-kt}$

(B)
$$k_{\Delta} > k_{C} > k_{R}$$

$$(\tilde{C}) k_B > k_A > k_C$$

(D)
$$k_{C} > k_{B} > k_{A}$$

Sol. Ans. (D)

38. A specific volume of H₂ requires 24 s to diffuse out of a container. The time required by an equal volume of O₂ to diffuse out under identical conditions, is

Sol.
$$\frac{r_{O_2}}{r_{H_2}} = \sqrt{\frac{M_{H_2}}{M_{O_2}}}; \quad \frac{r_{O_2}}{r_{H_2}} = \sqrt{\frac{2}{32}} \; \frac{r_{O_2}}{r_{H_2}} = \frac{1}{4} \qquad \qquad r_{O_2} : r_{H_2} = 1:4$$

Ans. (B)

39. Acetic acid reacts with sodium metal at room temperature to produce

(D) CO

2CH₃COOH + 2Na → 2CH₃COONa + H₂ ↑ Sol.

Ans. (B)

The equilibrium constant, K_c for $3C_2H_2(g) \rightleftharpoons C_6H_6(g)$ 40.

> is 4 L² mol⁻². If the equilibrium concentration of benzene is 0.5 mol L⁻¹, that of accetylene in mol L⁻¹ must be (A) 0.025 (B) 0.25(C) 0.05(D) 0.5

Sol.
$$K_c = \frac{[C_6H_6]}{[C_2H_2]^3}$$

$$4 = \frac{0.5}{[C_2 H_2]^3}$$

$$[C_2H_2]^3 = \frac{0.5}{4}$$

$$[C_2H_2]^3 = \frac{1}{8}$$

$$[C_2H_2] = \frac{1}{2}$$

$$[C_2H_2] = 0.5$$

Ans. (D)

41. The weight percent of sucrose (formula weight = 342 g mol⁻¹) in an aqueous solution is 3.42. The density of the solution is 1 g mL⁻¹, the concentration of sucrose in the solution in mol L⁻¹ is

- (C) 1.0

(D) 10

3.42 gm sacrose in 100 gm solution Sol. $d = 1 \text{ gm ml}^{-1}$

$$\therefore d = \frac{\text{mass}}{\text{volume}}$$

volume of solution =
$$\frac{100}{1}$$
 = 100ml

Molarity =
$$\frac{n}{v} \times 1000$$

Molarity =
$$\frac{3.42}{342 \times 100} \times 1000 = 0.1$$

Ans. (B)

- 42. The order of reactivity of K, Mg, Au and Zn with water is
 - (B) K > Mg > Zn > Au (C) K > Au > Mg > Zn (D) Au > Zn > K > Mg

(A) K > Zn > Mg > AuSol. $E_{red}^0 = K < Mg < Zn < Au$

Ans. (B)

43. Which of the following is an anhydride?

Ans. (A)

44. Which of the following metals will precipitate copper from copper sulphate solution?

- (C) Au

(D) Pt

 $E_{red}^0 = Sn < Cu < Hg < Au$ Sol.

The radii of the first Bohr orbit of H (r_H), He⁺ (r_{He} ⁺) and Li²⁺ (r_{Li} ²⁺) are in the order (A) r_{He} ⁺ > r_H > r_{Li} ²⁺ (B) r_H < r_{He} ⁺ < r_{Li} ²⁺ (C) r_H > r_{He} ⁺ > r_{Li} ²⁺ (D) r_{He} ⁺ < r_H < r_{Li} ²⁺ 45.

 $r = 0.529 \times \frac{n^2}{7}$ Sol.

 $r \alpha \frac{1}{2}$ So correct order is $r_H > r_{He}^+ > r_{Li}^{2+}$

BIOLOGY

46.	The Bowman's capsule, a part of the kidney is the site of (A) filtration of blood constituents (B) re-absorption of water and glucose (C) formation of ammonia (D) formation of urea					
Sol.	In Bowman's capsule ultrafiltration of blood occur. Ans. (A)					
47.	In human brain the sensation of touch, pain and temperature is controlled by the (A) parietal lobe of cerebrum (B) limbic lobe of cerebrum (C) temporal lobe of cerebrum (D) frontal lobe of cerebrum					
Sol.	Parietal lobe is sensory lobe for touch, pain temperature. Ans. (A)					
48.	A pathogan which can not be cultured in an artificial medium is, (A) protozoan (B) virus (C) becterium (D) fungus					
Sol.	Virus multiplies only in living cells Ans. (B)					
49. Sol.	Meiosis I and Meiosis II are characterised by the separation of, (A) homologous chromosomes; sister chromatids (B) sister chromatids; homologous chromosomes (C) centromere; telomere (D) telomere; centromere Meiosis I – Reduction Division (2n — → n), Separation of Homologous Chromosomes results in reduction of chromosome ploidy to half Meiosis II – Similar to mitosis sister chromatids separate					
	Ans. (A)					
50.	People suffering from albinism cannot synthesize (A) suberin (B) melanin (C) keratin (D) collagen					
Sol.	Melanin pigment synthesized from Tyrosine amino acid impart colour to skin. Ans. (B)					
51.	Short sightedness in humans can be corrected by using					
Sol.	(A) concave lens (B) convex lens (C) cylindrical lens (D) plain glass Image is formed before retina in short sightedness Ans. (A)					
52.	A person with blood group "A" can (a) donate blood to, and (b) receive blood from, (A) (a) persons with blood group "AB", and (b) persons with any blood group (B) (a) person with blood group "A" or "AB", and (b) "A" or "O" blood groups (C) (a) person with blood group "B" or "AB", and (b) "B" or "O" blood groups (D) (a) person with any blood group, and (b) "O" blood group only					
Sol.	O is universal donor; AB is universal recipient O blood group RBC donot have antigen AB blood group plasma donot have antibody Ans. (B)					

Sol.	(A) nucleosomes Mitochondria have circu Ans. (B)	(B) mitochondria	(C) peroxisomes	(D) lysosome		
54.	Which one of the following combinations is found in DNA? (A) Guanine and guanidine (B) Guanidine and cytosine					
Sol.	(C) Guanine and cytosine a Ans. (C)	ne are nitrogenous bases of	(D) Adenine and guanidine DNA.			
55.	Which one of the following is NOT a mode of asexual reproduction? (A) Binary fission (B) Multiple fission (C) Budding (D) Conjugation Conjugation involves transfer of DNA from one microbe to another.(e.g. Bacteria) Ans. (D)					
Sol.						
56.	Which one of the follow (A) Insects Ans. (A)	ring class of animals con (B) Fish	situtes the largest biomass (C) Mammals	on earth? (D) Reptilians		
57.	In the digestive system, the pH of the stomach and the intestine, respectively are,					
Sol.	(A) alkaline; acidic pH stomach = 1.5 – 2.5 Intestine = 7.4 – 7.6 Ans. (B)	(B) acidic; alkaline	(C) acidic; neutral	(D) acidic; acidic		
58.	The major nitrogenous excretory product in mammals is,					
Sol.	(A) amino acids NH ₃ is converted to ure Ans. (C)	(B) ammonia a in hepatocytes.	(C) urea	(D) uric acid		
59.	Which of the following plant traits (characters) is NOT an adaptation to dry (Xeric) habitats? (A) Sunken stomata on leaves (B) Highly developed root system (C) Thin epidermis without a cuticle on stem and leaves (D) Small leaves and photosynthetic stem					
Sol.	Thick cuticle prevent los Ans. (C)	ss of water				
60.	Biological diversity increases with the productivity of an ecosystem. In which of the following habitats do visee the greatest diversity of species? (A) Tropical dry grasslends (B) Temperate deciduous forests (C) Alpine grasslends (D) Tropical evergreen forests Ans. (D)					



PART-II Two Mark Questions

MATHEMATICS

- 61. Let a, b, c, d, e be natural numbers in an arithmetic progression such that a + b + c + d + e is the cube of an integer and b + c + d is square of an integer. The least possible value of the number of digits of c is

 (A) 2 (B) 3 (C) 4 (D) 5
- **Sol.** a = C 2D

$$b = C - D$$

$$c = C$$

$$d = C + D$$

$$e = C + 2D$$

$$a + b + c + d + e = 5c = \lambda^3$$

$$b + c + d = 3c = \mu^2$$

$$\Rightarrow 3\lambda^3 = 5\mu^2$$

$$\frac{\lambda^3}{5} = \frac{\mu^2}{3}$$
 least possibility

$$\lambda = 5 \times 3$$
, $\mu = 5 \times 3 \times 3$

$$\lambda = 15 \ \mu = 45$$

$$C = \frac{(45)^2}{3} = 15 \times 45 = 675$$

Ans. (B)

- 62. On each face of a cuboid, the sum of its perimeter and its area is written. Among the six numbers so written, there are three distinct numbers and they are 16, 24 and 31. The volume of the cuboid lies between
- (A) 7 and 14
- (B) 14 and 21
- (C) 21 and 28
- (D) 28 and 35

- Sol. 2(a + b) + ab = 162(b + c) + bc = 24
- ... (1)
- 2(c + a) + ca = 31

- <u>2)</u> 3)
- From equation (2), equation (3) \Rightarrow (a b) (2 + c) = 7 From equation (2) and equation (4) \Rightarrow 4a = 2 + 5b

... (4)

- Solve equation (1) and (5)
- b = 2 a = 3, c = 5
- Volume = 30

Ans. (D)

63. Let ABCD be a square and let P be point on segment CD such that DP : PC = 1 : 2. Let Q be a point on segment AP such that \angle BQP = 90°. Then the ratio of the area of quadrilateral PQBC to the area of the square ABCD is

(A)
$$\frac{31}{60}$$

(B)
$$\frac{37}{60}$$

(C)
$$\frac{39}{60}$$

(D)
$$\frac{41}{60}$$
 3

Sol.

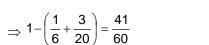
$$CP = \frac{2}{3}a$$

$$PD = \frac{a}{3}$$

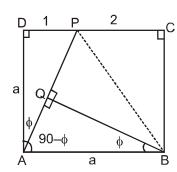
$$tan\phi = \frac{1}{3} (In \Delta APD)$$

Now,
$$\angle DAP = \angle QBA = \phi$$

Required ratio =
$$\frac{\text{area of PQBC}}{\text{a}^2} = \frac{\text{a}^2 - \left(\text{area of } \Delta \text{ADP} + \text{area of } \Delta \text{AQB}\right)}{\text{a}^2}$$



Ans. (D)



Suppose the height of a pyramid with a square base is decreased by p% and the lengths of the sides of its 64. square base are increased by p% (where p > 0). If the volume remains the same, then

(A) 50

- (B) 55
- (C) 60
- (D) 65

Sol.
$$\frac{1}{3}(x^2)y = \frac{1}{3}x^2\left(\frac{100+p}{100}\right)^2\left(\frac{100-p}{100}\right)$$
$$\Rightarrow p^2 + 100p - 100^2 = 0$$
$$p = \sqrt{12500} - 50$$
$$60 Ans. (C)$$

- 65. There are three kinds of liquids X, Y, Z,. Three jars J_1 , J_2 , J_3 contain 100 ml of liquids X, Y, Z, respectively. By an operation we mean three steps in the following order:
 - stir the liquid in J_1 and transfer 10 ml from J_1 into J_2 ; stir the liquid in J_2 and transfer 10 ml from J_2 into J_3 ; stir the liquid in J_3 and transfer 10 ml from J_3 into J_4 ;

After performing the operation four times, let \dot{x} , \dot{y} , \dot{z} be the amounts of \dot{X} , \dot{Y} , \dot{Z} , respectively, in \dot{J}_1 . Then (A) x > v > z(B) x > z > y(C) y > x > z(D) z > x > v

Sol. After one operation amount of x, y, z in J, respectively are

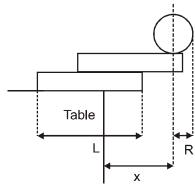
$$90 + 10 \times \left(\frac{1}{11}\right)^2$$
, $100\left(\frac{1}{11}\right)^2$, $\frac{100}{11}$

Similarly we can find after four operations amount of x,y,z in J_4 .

Ans. (B)

PHYSICS

66. Two identical uniform rectangular blocks (with longest side L) and a solid sphere of radius R are to be balanced at the edge of a heavy table such that the centre of the sphere remains at the maximum possible horizontal distance from the vertical edge of the table without toppling as indicated in the figure.



If the mass of each block is M and of the sphere is M/2, then the maximum distance x that can be achieved

(B) 5L/6

(C)(3L/4 + R)

(D)
$$(7L/15 + R)$$

Sol. 2 + S System lie above edge of 1.

$$\frac{M}{2}y - M\left(\frac{L}{2} - y\right) = 0$$

$$\frac{y}{2} + y = \frac{L}{2}$$

$$y = \frac{L}{3}$$

Now 1 + 2 + S centre of mass will lie above the table

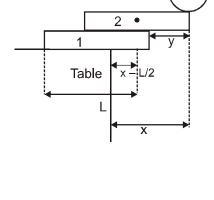
$$\frac{3M}{2}\left(x-\frac{L}{3}\right)+M\left(x-\frac{L}{3}-\frac{L}{2}\right)=0$$

$$\frac{3x}{2} - \frac{L}{2} + x - \frac{L}{3} - \frac{L}{2} = 0$$

$$\frac{5x}{2} = \frac{4L}{3}$$

$$x = \frac{8L}{15}$$

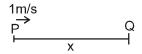
Ans.



- **67.** Two skaters P and Q are skating towards each other. Skater P throws a ball towards W every 5 s such that it always leaves her hand with speed 2 ms⁻¹ with respect to the ground. Consider two cases:
 - (I) P runs with speed 1 ms⁻¹ towards Q while Q remains stationary
 - (II) Q runs with speed 1 ms⁻¹ towards P while P remains stationary.

Note that irrespective of speed of P, ball always leaves P's hand with speed 2 ms⁻¹ with respect to the ground. Ignore gravity. Balls will be received by Q

- (A) one every 2.5 s in case (I) and one every 3.3 s in case (II)
- (B) one every 2 s in case (I) and one every 4 s in case (II)
- (C) one every 3.3 s in case (I) and one every 2.5 s in case (II)
- (D) one every 2.5 s in case (I) and one every 2.5 s in case (II)
- **Sol.** (I) Let initial distance between P & Q is x



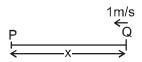
at $t_1 = \frac{x}{2}$ a receive the ball.

Next ball

$$t_2 = \frac{x-5}{2} + 5$$

$$\Delta t = \frac{5}{2}$$

(II) in second case at t = 0 P throws the ball.



$$t_1 = \frac{x}{3}.$$

Next ball

$$t_2 = \frac{x-5}{3} + 5$$

$$\Delta t = \frac{10}{3}$$

Ans. (A)

A 10.0 W electrical heater is used to heat a container filled with 0.5 kg of water. It is found that the temperature of the water and the container rise by 3 K in 15 minutes. The container is then emptied, dried, and filled with 2 kg of an oil. It is now observed that the same heater raises the temperature of the container-oil system by 2 K in 20 minutes. Assuming no other heat losses in any of the processes, the specific heat capacity of the oil is

(A)
$$2.5 \times 10^3 \, \text{JK}^{-1} \, \text{kg}^{-1}$$
 (B) $5.1 \times 10^3 \, \text{JK}^{-1} \, \text{kg}^{-1}$ (C) $3.0 \times 10^3 \, \text{JK}^{-1} \, \text{kg}^{-1}$ (D) $1.5 \times 10^3 \, \text{JK}^{-1} \, \text{kg}^{-1}$

Sol. Pt =
$$m_w S_w \Delta T + m_c s_c \Delta T$$

 $10 \times 15 \times 60 = 0.5 \times 4200 \times 3 + m_c s_c \times 3$
 $9000 = 6300 + m_c s_c 3$

$$m_c s_c = 900 \text{ J/k}.$$

Now, for oil

$$10 \times 20 \times 60 = 2 \times S_0 \times 2 + 900 \times 2$$

$$12000 - 1800 = 4 S_0$$

$$S_0 = \frac{10200}{4} = 2.51 \times 10^3 \text{ J/kg-k}$$

Ans. (A)

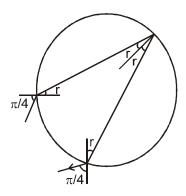
69. A ray of light incident on a transparent sphere at an angle $\pi/4$ and refracted at an angle r, emerges from the sphere after suffering one internal reflection. The total angle of deviation of the ray is

(A)
$$\frac{3\pi}{2} - 4r$$

(B)
$$\frac{\pi}{2} - 4r$$
 (C) $\frac{\pi}{4} - r$

(C)
$$\frac{\pi}{4}$$
 – I

(D)
$$\frac{5\pi}{2} - 4r$$



Sol.

$$S_1 = \frac{\pi}{4} - r$$

$$S_2 = \pi - 2r$$

$$S_3 = \frac{\pi}{4} - r$$

$$S = S_1 + S_2 + S_3 = \frac{3\pi}{2} - 4r$$

Ans. (A)

- 70. An electron with an initial speed of 4.0 × 10⁶ ms⁻¹ is brought to rest by an electric field. The mass and charge of an electron are 9 × 10⁻³¹ kg and 1.6 × 10⁻¹⁹C, respectively. Identify the correct statement
 - (A) The electron moves from a region of lower potential to higher potential through a potential difference of 11.4 μV.
 - (B) The electron moves from a region of higher potential to lower potential through a potential difference of
 - (C) The electron moves from a region of lower potential to higher potential through a potential difference of 45
 - (D) The electron moves from a region of higher potential to lower potential through a potential difference of 45

Sol.

$$qV = \frac{1}{2} mv^2$$

$$V = \frac{1}{2} \frac{mv^2}{q}$$

$$V = \frac{1}{2} \times \frac{9 \times 10^{-31} \times (4 \times 10^6)^2}{1.6 \times 10^{-19}} = 45V$$

45 V from higher to lower potential.

Ans. (D)

CHEMISTRY

71. The degree of dissociation of acetic acid (0.1 mol L⁻¹) in water (K_a of acetic acid is 10⁻⁵) is

Sol.

$$C = 0.1 M$$

$$K_a = \alpha^2 C$$

$$K_a = 10^{-5}$$

$$10^{-5} = \alpha^2 \times 0.1$$

$$10^{-5} = \alpha^2 \times 0.1$$

$$\alpha^2 = 10^{-4}$$

$$\alpha = 10^{-2}$$

Ans. (A)

72. Compound 'X' on heating with Zn dust gives compound 'Y' which on treatment with O₃ followed by reaction with Zn dust gives propionaldehyde. The structure of 'X' is

$$(A) \xrightarrow{Br} (B) \xrightarrow{Br} (C) \xrightarrow{Br} (D) \xrightarrow{Br}$$

Ans. (C)

73. The amount of metallic Zn (Atomic weight = 65.4) required to react with aqueous sodium hydroxide to produce 1 g of H_2 , is

(A) 32.7 g (B) 98.1 g (C) 65.4 g (D) 16.3 g
Sol.
$$Zn + 2OH^- + 2H_2O \longrightarrow [Zn (OH_4)]^{2-} + H_2$$

 $2gm (1 mole) H_2 is given by 65.4 gm of Zn$

1 gm is given by $\frac{65.4}{2}$ gm of Zn

Ans. (A)

74. Natural abundances of 12 C and 13 C isotopes of carbon are 99% and 1%, respectively. Assuming they only contributes to the mol. wt. of C_2F_4 , the percentage of C_2F_4 having a molecular mass of 101 is (A) 1.98 (B) 98 (C) 0.198 (D) 99

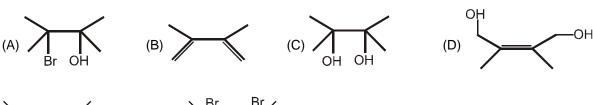
Sol. Percentage of
$$C_2F_4$$
 of Molar mass $100 = \frac{1}{100} \times \frac{1}{100} \times 100 = 0.01\%$

Percentage of C_2F_4 of Molar mass $102 = \frac{99}{100} \times \frac{99}{100} \times 100 = 98.01\%$

Percentage of C_2F_4 of Molar mass 101 = 100 – (0.01 + 98.01) = 1.98%

Ans. (A)

75. 2, 3-Dimethylbut-2-ene when reacted with bromine forms a compound which upon heating with alcoholic KOH produce the following major product.



Ans. (B)

BIOLOGY

- **76.** Sister chromatids of a chromosome have
 - (A) different genes at the same locus
 - (B) different alleles of the same gene at the same locus
 - (C) same alleles of the same gene at the same locus
 - (D) same alleles at different loci
- **Sol.** Sister chromatids are identical DNA molecules (In somatic cells)

Ans. (C)

- 77. A diabetic individual becomes unconscious after self-administering insulin. What should be done immediately to revive the individual?
 - (A) Provide him sugar
 - (B) Give him high dose of insulin
 - (C) Provide him salt solution
 - (D) Provide him lots of water
- **Sol.** Insulin lowers blood sugar level and in this case brain is getting inadequate sugar/ glucose

Ans. (A)

- **78.** A regular check on the unborn baby of a lady towards the end of her pregnancy showed a heart rate of 80 beats per minute. What would the doctor infer about the baby's heart condition from this?
 - (A) Normal heart rate
- (B) Faster heart rate
- (C) Slower heart rate
- (D) Defective brain function

Sol. Infants have higher heart rate.

Ans. (C)

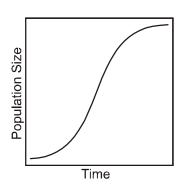
79. Three uniformly watered plants i, ii and iii were kept in 45% relative humidity, 45% relative humidity with blowing wind and 95% relative humidity, respectively. Arrange these plants in the order (faster to slowest) in which they will dry up.

(A)
$$i = ii$$
, iii

(D)
$$iii$$
, $i = ii$

Ans.(B)

80. Many population colonising a new habitat show a logistic population growth pattern over time, as shown in the figure below.



In such a population, the POPULATION growth rate

(A) stays constant over time

(B) increases and then reaches an asymptote

(C) decreases over time

- (D) increases to a maximum and then decrease
- **Sol.** $\frac{dN}{dt} = rN\left(1 \frac{N}{k}\right)$, As N approach k $\frac{dN}{dt}$ decrease that is asymptote.

Ans. (B)

* * * * *