



Address:JB 20, Near Jitendra Cinema, City Centre, Sec 4, Bokaro Mobile: 7488044834 Website: www.vidyadrishti.org

## Notes for School Exams Physics XI

## **Fluid Mechanics**

## **Mechanical Properties of Matter**

#### Author: P. K. Bharti (B. Tech., IIT Kharagpur)

H. O. D. Physics, Concept Bokaro Centre

Mb: 7488044834

© 2007 P. K. Bharti

## <mark>20</mark>13-2015

Prerequisite: Laws of motion (F. B.D.)	Introduction
Content <ul> <li>Introduction</li> <li>Fluids</li> <li>Density &amp; specific volume</li> <li>Specific weight &amp; specific gravity</li> <li>Compressibility &amp; viscosity</li> <li>Ideal fluid</li> </ul>	<ul> <li>A fluid is any substance which flows because its particles are not rigidly attached to one another. This includes liquids, gases and even some materials which are normally considered solids, such as glass.</li> <li>Fluid mechanics is the study of fluids either in motion (<i>fluid dynamics</i>) or at rest (<i>fluid statics</i>) and the subsequent effects of the fluid upon the boundaries, which may be either solid surfaces or interfaces with other fluids.</li> </ul>
Hydrostatics	Fluids
<ul> <li>Gauge pressure &amp; vacuum pressure</li> <li>Mercury barometer</li> <li>Manometer</li> <li>Pascal's Law</li> </ul>	<ul> <li>From the point of view of fluid mechanics, all matter consists of only two states, fluid and solid. Both liquids and gases are fluids.</li> </ul>
<ul> <li>Hydraulic lift</li> <li>Buoyant force</li> <li>Archimedes' Principle</li> <li>Fluids in motion</li> <li>Reynolds number</li> <li>Equation of continuity</li> <li>Bernoulli's Principle</li> <li>Torricelli's theorem (speed of efflux)</li> <li>Venturimeter</li> <li>Viscosity</li> </ul>	<ul> <li>Thus all The technical distinction lies with the reaction of the two to an applied shear or tangential stress. A solid can resist a shear stress by a static deformation; a fluid cannot. Any shear stress applied to a fluid, no matter how small, will result in motion of that fluid. The fluid moves and deforms continuously as long as the shear stress is applied.</li> <li>Hence, we can define <i>fluid as substance which cannot resist any shear stress applied to it.</i></li> </ul>
<ul> <li>Coefficient of viscosity</li> <li>Stokes law</li> </ul>	Donsity & Specific Volume
<ul> <li>Terminal velocity</li> <li>Terminal velocity of a small sphere</li> <li>Surface tension</li> <li>Reason for surface tension</li> <li>Definition of surface tension</li> <li>Surface energy</li> <li>Excess pressure inside a soap bubble</li> <li>Capillary rise</li> </ul>	<ul> <li>Density &amp; Specific Volume</li> <li>Density: The mass per unit volume of material is called the density, which is generally expressed by the symbol ρ.</li> <li>ρ = m/V</li> <li>The SI unit of density is kg/m<sup>3</sup>.</li> <li>The density of water at 4°C and 1 atm (101325 Pa, 3)</li> </ul>
	standard atmospheric pressure) is $1000 \text{ kg/m}$ or $1\text{g/cm}$ .
<ul> <li>NOTE: I have followed NCERT Physics book to prepare this notes. Even IIT-JEE (now JEE Advanced) follows NCERT. Please refer NCERT whenever two books disagree on a particular topic. All the best for Boards, JEE Main &amp; JEE Advanced.</li> <li>Pranjal K. Bharti H. O. D. Physics at Concept Bokaro Centre</li> </ul>	<ul> <li>The density of mercury is taken as 13600 kg/m.</li> <li>Specific volume: The reciprocal of density, i.e. the volume per unit mass, is called the specific volume. specific volume = 1/ρ = V/m</li> <li>SI Unit : m<sup>3</sup>/kg</li> </ul>
B. Tech., I.I.T. Kharagpur	Specific Weight & Specific Gravity
Mb: 7488044834 Email: <u>pkbharti.iit@gmail.com</u> Website: www.vidyadrishti.org	<ul> <li>Specific weight: The specific weight of a fluid, denoted by γ, is its weight per unit volume.</li> <li>Therefore, γ = mg/V = ρg     </li> </ul>

Р

Specific gravity or relative density (Important): Specific gravity, denoted by SG or RD, is the ratio of a fluid density to a standard reference fluid, water (for liquids), and air (for gases):

density of liquid Relative density of a liquid = density of water at 4°C

$$\mathrm{SG}_{liquid} = rac{
ho_{liquid}}{
ho_{water}}$$

- Similarly, SG<sub>gas</sub> =  $\frac{\rho_{gas}}{\rho_{air}}$
- For example, the specific gravity of mercury (Hg) is  $SG_{Hg} = 13,600/1000 \approx 13.6.$

#### Exercise 1

- Find the dimensions of a) density and b) specific gravity. 1.
- Find the specific density of a liquid whose density is a) 8300 kg/m<sup>3</sup> and b) 2 g/cm<sup>3</sup>. 2.
- 3. If the relative density of a fluid is 11.1 find its density in S.I. and CGS units.

#### Answers

- 1. a)  $ML^{-3}T^{0}$  b)  $M^{0}L^{0}T^{0}$
- 2. a) 8.3 b) 2
- 11100 kg/m , 11.1 g/cm

#### **Compressibility & viscosity**

- Compressibility: Compressibility is the measure of the change in volume a substance undergoes when a pressure is exerted on the substance.
- Liquids are generally considered to be *incompressible*, meaning density of the liquid is independent of the variation in pressure and always remains constant.
- **Viscosity:** *Viscosity is a fluid property that measures the* resistance of the fluid to deforming due to a shear force.
- Viscosity is the internal friction of a fluid which makes it resist flowing past a solid surface or other layers of the fluid. Viscosity can also be considered to be a measure of the resistance of a fluid to flowing. A thick oil has a high viscosity; water has a low viscosity.

#### **Ideal Fluid**

- An ideal fluid is one that is incompressible and has no viscosity.
- Ideal fluids do not actually exist. We are going to assume given fluid to be ideal, meaning incompressible and nonviscous fluid unless or otherwise stated.

#### Pressure

- 1. Pressure (P) is the force per unit area applied on a surface in a direction **perpendicular** to that surface.
- 2. We can classify pressure in two categories: Average pressure and pressure at a point.
- 3. Average Pressure: Suppose a force F is applied to a surface of area A. Then, average pressure is defined as





where  $F_{\perp}$  is the component of force F perpendicular to the surface.

4. **Pressure at a point:** Suppose a infinitesimally small surface area dA centred at a point. Suppose a force  $dF_{\perp}$ acts perpendicular to this force at that surface dA. Then, pressure at that point is given by



5. Pressure in a fluid at a particular point acts equally in all direction.



Here 
$$P_1 = P_2 = P_3 = P_4 = P_5$$

#### Important point about pressure

S.I unit of pressure is N/m called Pascal (Pa).

$$1 Pa = 1 N/m$$

- Pressure is a scalar quantity (at school level).
- 6. Pressure in a fluid at a particular point acts equally in all direction.
- The atmosphere (atm) exerts certain pressure at a point depending on the column or height of atmosphere lying above that point. The average pressure of the atmosphere at sea level is known as atmospheric pressure (atm).
- The **atmospheric pressure** is about  $1.013 \times 10^5$  Pa and is denoted by  $P_0$ . Thus,  $P_0 = 1.013 \times 10^5 \text{ Pa}$
- This gives us another unit for pressure, the atmosphere
- (atm), where  $1 \text{ atm} = 1.013 \times 10 \text{ Pa}$
- There is one more unit prevalent in laboratory, known as

Bar. 1 Bar = 10 Pa

#### Hydrostatics

• Hydrostatics is about the pressures exerted by a **fluid at rest.** Any fluid is meant, not just water.

#### Pressure variation with depth in Hydrostatic

- Let us consider an **incompressible fluid** of **uniform density**  $\rho$  at **rest**.
- Consider an imaginary fluid volume (a cube, each face having area A) at rest. The sum of all the forces on this volume must be zero as it is in equilibrium.
- There are three vertical forces:
- ✓ The weight:  $mg = \rho V g = \rho h A g$
- ✓ The upward force from the pressure  $P_2$  on the bottom surface:  $F_2 = P_2 A$
- ✓ The downward force from the pressure  $P_1$  on the top surface:  $F_1 = P A$

i. Therefore, at equilibrium, we have,  

$$F_2 - F_1 - mg = 0$$

$$\Rightarrow P_A - P_A - mg = 0$$
  
$$\Rightarrow (P_2 - P_1) A - \rho h A g = 0$$
  
$$\Rightarrow (P_2 - P_1) - \rho h g = 0$$



- $\Leftrightarrow \frac{P_2 P_1}{P_2 P_1} = \rho g h$  $\Leftrightarrow \frac{P_2 P_1}{P_2 P_1} = \rho g h$
- Therefore, pressure  $P_2$  at depth *h* is  $\rho gh$  greater than pressure *P*.
- Hence, pressure increases with depth.

#### **Important Points in Hydrostatics**

1. Pressure in a continuously distributed uniform, incompressible static fluid varies only with vertical distance and is independent of the shape of the container. The pressure increases with depth in the fluid. Pressure  $P_2$  at depth *h* is  $\rho gh$  greater than pressure  $P_1$ . Thus,

 $P_2 - P_1 = \rho g h$ 

Taking variation in acceleration due to gravity with depth

or height into account we have,  $\frac{dp}{dh} = -\rho g$ 

2. Pressure in a fluid at a particular point acts equally in all direction.



- 3. The pressure is the same at all points on a given horizontal plane in the fluid.
- An illustration of this is shown in the given Fig. The free surface of the container is atmospheric and forms a horizontal plane. Points *A*, *B*, *C*, and *D* are at equal depth in a horizontal plane and are interconnected by the same fluid; therefore all points have the same pressure. The same is true of points *P*, *Q*, and *R* which all have the same lower pressure than at *A*, *B*, *C*, and *D*. However, point *S*, although at the same depth as *A*, *B*, and *C*, has a different pressure because it lies beneath a different fluid. Please note that points *T* and *U* have same pressure.



- 4. Forces acting on a fluid in equilibrium have to perpendicular to its surface, because it cannot sustain the shear stress.
- 5. **Free body diagram of a liquid:** Forces on a fluids in equilibrium are (neglecting viscous forces) are:
  - Weight *mg* in downward direction
  - Force  $P_0A_1$  from atmospheric pressure in downward direction
  - Normal force  $(P_0+\rho gh)A_2$  from bottom surface in upward direction (how?)

All of these forces are in vertical direction. By calculation it is found that net force is not zero. It means there must be some force in the vertical direction to maintain the equilibrium. Where does this force comes from? Yes! This force is due to walls of the container. Hence there is a fourth force:

• Force  $F_{\text{wall}}$  from walls of the container in the vertical (say upward) direction.



#### 6. **Pressure difference in an accelerating fluids:**

Consider a liquid kept at rest in a beaker as shown in figure. In this case we know that pressure do not change in horizontal direction (x-direction), it decreases upward along y-direction. So, we can write the equations,



But, suppose the beaker is accelerated and it has components of acceleration  $a_x$  and  $a_y$  in x and y directions respectively, then the pressure decreases along both x and y direction. The above equation in that case reduces to,

$$\frac{dp}{dx} = -\rho a_x \quad \text{and} \quad \frac{dp}{dy} = -\rho \left(g + a_y\right)$$

7. Free surface of a liquid accelerated in horizontal direction:

Consider a liquid placed in a beaker which is accelerating

horizontally with an acceleration *a* . Then,  $\tan \theta = \frac{a}{a}$ 

Proof: Consider a fluid particle of mass m at point P on the surface of liquid. From the accelerating frame of reference, two forces are acting on it,

(i) pseudo force (ma)(ii) weight (mg)



• Net force in equilibrium should be perpendicular to the surface.

$$\therefore \tan \theta = \frac{ma}{mg}$$

а

g

or 
$$\tan \theta =$$

#### **Gauge Pressure and Vacuum Pressure**

- Absolute pressure (P): The pressure at a point is known as absolute pressure.
- Gauge Pressure (when  $P > P_o$ ): The excess pressure above atmospheric pressure is called as gauge pressure. Therefore,

Gauge Pressure =  $P - P_o$ 

• Vacuum Pressure (when  $P < P_0$ ) Vacuum Pressure = Atmospheric Pressure – Absolute Pressure

$$\Leftrightarrow$$
 Vacuum Pressure =  $P_o - P$ 

#### **The Mercury Barometer**

shows a very basic mercury barometer, • Figure a device used to measure the pressure of the atmosphere. The long glass tube is filled with mercury and inverted with its open end in a dish of mercury, as the figure shows. The space above the mercury column contains only mercury vapor, whose pressure is so small at ordinary temperature that it can be neglected.

$$P_2 - P_1 = \rho g h$$

$$\Leftrightarrow P_{o} = \rho g h$$



• The atmospheric pressure is often given as the length of mercury column in a barometer. Thus, a pressure of 76cm of mercury means, 1 atmospheric pressure.

where  $\rho = \text{density of the mercury.}$ 

#### Manometer

• *Manometer is a simple device to measure the pressure in a closed vessel containing a gas.* It consists of a U-shape tube having some liquid. One end of the tube is open to the atmosphere and the other end is connected to the vessel as shown in figure. The pressure of the gas is equal to P<sub>1</sub>. From hydrostatic,

$$P_2 - P_1 = \rho g h$$
$$\Leftrightarrow P_{gas} - P_o = \rho g h$$

$$\Leftrightarrow P_{\rm gas} = P_{\rm o} + \rho g h$$



where  $P_{\text{gas}} = \text{ pressure of the gas}$ 

 $P_{\rm o}$  = the atmospheric pressure

h = difference in levels of the liquid in the two arms

 $\rho$  = the density of the liquid.

#### Pascal's Law

- Pascal's law states that "if a pressure is applied to an *enclosed fluid*, it is transmitted *undiminished* to every portion of the fluid and the walls of the containing vessel."
- Applications of Pascal's law: Hydraulic lift Hydraulic brakes Cycle pump

#### Hydraulic Lift

- A hydraulic lift uses Pascal's principle. Hydraulic lift is used to raise heavy loads such as car. It contains of two vertical cylinders A and B of different cross sectional areas A and A. Pistons are fitted in both the cylinders as shown in fig.
- A small force is applied  $F_1$  to a small piston of area  $A_1$ and cause a pressure increase on the fluid.
- According to Pascal's Law this increase in pressure P is transmitted to the larger piston of area A and the fluid
- exerts a force  $F_2$  on this piston.
- Thus, from Pascal's Law  $P = F_{I} / A_{I} = F_{2} / A_{2}$   $\Leftrightarrow F = F (A / A)$
- Thus if  $A_2 >> A_1$ , even a small force  $F_1$  is able to generate a large force  $F_2$  which can raise the load.

#### **Buoyant Force**

- When an object is fully or partially submersed in a fluid, the surrounding fluid exerts a *net upward force* which is known as the **buoyant force** or **upthrust**.
- It is easier to lift a bucket immersed in water because of buoyant force.
- NOTE:
- 1. The buoyant force comes from the **pressure** exerted on the object by the surrounding fluid.
- When showing F.B.D., we need to show either buoyant force in the upward direction or forces due to pressure. We never show both buoyant force and forces due to pressure in the same F.B.D.

#### Archimedes Principle (Buoyant Forces)

Archimedes' Principle states that a body which is completely or partially submerged in a fluid experiences a net upward force called the buoyant force, B, which is equal in magnitude to the weight of the fluid displaced by the object. Thus,

Buoyant force = weight of the displaced liquid

$$B = V_{\text{im}} \rho_{\text{liquid}} g$$
 (Buoyant force)

where,

B = magnitude of Buoyant force

 $V_{\rm im}$  = volume of displaced liquid = immersed volume of solid

 $\rho_{liquid}$  = density of liquid

#### **Proof:**

 As shown in Fig., consider a body of height *h* lying inside a liquid of density ρ,. Area of cross-section of the body is *A*. The forces on the sides of the body cancel out.

 $F_1$ 

mg

 $P_2$ 

- There are two vertical forces due to pressures:
- ✓ The upward force from the pressure  $P_2$  on the bottom surface:

$$F_2 = P_2 A$$

 $\checkmark$  The downward force from the pressure  $P_{1}$  on the top surface:

$$F_{\perp} = P$$

• The resultant force  $(F_2 - F_1)$  is acting on the body in the upward direction and is called upthrust or buoyant force (B).

$$\therefore \mathbf{B} = F_2 - F_1 = \mathbf{P}_2 \mathbf{A} - \mathbf{P}_1 \mathbf{A} = (\mathbf{P}_2 - \mathbf{P}_1)\mathbf{A} = h\rho g \mathbf{A}$$

 $(:: P_2 - P_1 = \rho gh)$ 

• But *Ah* = *V*, the volume of the body = volume of liquid displaced

 $\therefore B = V\rho g = Mg$ 

(::  $M = V\rho$  = mass of liquid displaced)

• This proves the Archimedes' principle.

#### Law of floatation

• Consider an object of volume V and density  $\rho_{solid}$  floating in a liquid of density  $\rho_{liquid}$ . Let  $V_{im}$  be the volume of object immersed in the liquid. For equilibrium of the object,

Weight = upthrust  $V \rho_{\text{solid}} g = V_{\text{im}} \rho_{\text{liquid}} g$ 

$$\Leftrightarrow \left| \begin{array}{c} rac{V_{im}}{V} = rac{
ho_{solid}}{
ho_{liquid}} & ext{(fraction of volume immersed in liquid)} \end{array} 
ight.$$

- This is the fraction of volume immersed in liquid.
- Three possibilities may arise:
  - i.  $\rho_{solid} < \rho_{liquid}$ : Body is *partially submerged* in liquid. The fraction submerged is given by the relation

$$\frac{\mathbf{v}_{im}}{V} = \frac{\rho_{solid}}{\rho_{liquid}}$$

ii.  $\rho_{\text{solid}} = \rho_{\text{liquid}}$ : Body is *completely submerged* in liquid. Body remains floating in liquid.

iii. 
$$\rho_{\text{solid}} > \rho_{\text{liquid}}$$
: Body will *sink* in liquid.

•

streamlines

#### Apparent Weight

If an object is placed inside a fluid then, **Apparent weight = (Actual weight) – (Buoyant force)** 

#### **Buoyant Force in Accelerating Fluids**

• Suppose a body is dipped inside a liquid of density  $\rho_{liauid}$ 

placed in an elevator moving with an acceleration  $\vec{a}$ . The buoyant force F in this case becomes,

$$F = V_{im} \rho_{liquid} g_{eff}$$

Here,  $g_{eff} = \left| \vec{g} - \vec{a} \right|$ 

- Concept of  $g_{eff}$  is explained in chapter Simple Harmonic Motion.
- For example, if the lift is moving upwards with an acceleration a, the value of  $g_{eff}$  is g + a and if it is moving downwards with acceleration a, the  $g_{eff}$  is g a. In a freely falling lift  $g_{eff}$  is zero (as a = g) and hence, net buoyant force is zero. This is why, in a freely falling vessel filled with some liquid, the air bubbles do not rise up (which otherwise move up due to buoyant force).

#### Space for notes:

#### Fluids in Motion

- All fluid flow is classified into one of two broad categories or regimes. These two flow regimes are laminar flow and turbulent flow.
- Laminar Flow: Laminar flow is also referred to as streamline or viscous flow or steady flow. When a liquid flows such that each particle of the passing a given point moves along the same path and has the same velocity as its predecessor, the flow is called streamline flow or steady flow.
- **Streamline:** A streamline may be defined as the path, the tangent to which at any point gives the direction of the flow of liquid at that point.
- **Tube of flow.** A bundle of streamline forming a tubular region is called a tube of flow.



- **Turbulent Flow:** Turbulent flow is characterized by the irregular movement of particles of the fluid. The particles travel in irregular paths with no observable pattern and no definite layers.
- **Critical velocity.** The critical velocity of a liquid is that limiting (maximum) value of its velocity of flow upto which the flow is streamlined and above which the flow becomes turbulent.

#### **Reynolds Number**

- The Reynolds number is a dimensionless number comprised of the physical characteristics of the flow. The flow regime (either laminar or turbulent) is determined by evaluating the Reynolds number of the flow .
- The Reynolds number for fluid flow is given by

$$R_e = \frac{\rho v D}{\eta}$$
 (Reynolds number)

where

 $R_e$  = Reynolds number (dimensionless; have not any unit)

- v = average velocity
- D = diameter of pipe
- $\eta$  = viscosity of fluid (to be studied later)
- $\rho$  = fluid density
- Important point to note about Reynolds number:
  - i. For practical purposes (as per NCERT), if the Reynolds number is less than 1000, the flow is laminar. If it is greater than 2000, the flow is turbulent.
  - ii. Flows with Reynolds numbers between 1000 and 2000 are sometimes referred to as unsteady flows.
  - iii.  $R_e$  represents the ratio of inertial force (force due to inertia i.e. mass of moving fluid or due to inertia of obstacle in its path) to viscous force.

#### **Equation of Continuity**

- Equation of continuity states that total mass of fluids going into the tube through any cross-section should be equal to the total mass coming out of the same tube from any other cross section in the same time. The continuity equation results from conservation of mass.
- Let us consider mass is entering with speed  $v_1$  at left end and flowing out with speed  $v_2$ .



• Clearly, in a time interval  $\Delta t$ , mass entering = (mass per unit time) × time =  $\rho A_1 v_1 \Delta t$ .

(Hint: mass = 
$$\rho V = \rho A l$$

Hence, mass per unit time =  $m/t = \rho A l/t = \rho A v$ 

- And, in same time interval  $\Delta t$ , mass leaving
  - = (mass per unit time) × time

$$= \rho A_2 v_2 \Delta t$$

• Hence, from conservation of mass we have,  $\rho A_{1} v_{1} \Delta t = \rho A_{2} v_{2} \Delta t$ 

 $\Leftrightarrow A_{v} = A_{v}$ (Equation of Continuity)

• The product of the area of cross section and the speed remains the same at all points of a tube of flow. This is called the "equation of continuity" and expresses the law of conservation of mass in fluid dynamics.

#### **Bernoulli's Principle**

- Bernoulli's Principle relates the speed of a fluid at a point the pressure at that point and the height of that point above a reference level. It is just the application of workenergy theorem in the case of fluid flow.
- We here consider the case of *irrotational and steady flow* of an *incompressible* and *non viscous liquid*.
- Bernoulli's Principle states that the sum of pressure energy per unit volume, kinetic energy per unit volume and potential energy per unit volume of an incompressible, non-viscous fluid in a streamlined irrotational flow remains constant along a streamline. Mathematically,

$$p + \frac{1}{2}\rho v^2 + \rho gh = \text{constant}$$
$$\Leftrightarrow p_1 + \frac{1}{2}\rho v_1^2 + \rho gh_1 = p_2 + \frac{1}{2}\rho v_2^2 + \rho gh_2$$

#### **Proof of Bernoulli's Principle**

- One end of the pipe is horizontal at a height  $h_1$  above some reference level and has uniform cross-sectional area  $A_1$  up to some length. The other end is at a height  $h_2$  from the reference level and has uniform cross-sectional area  $A_2$ .
- Now consider the portion of fluid shown by shaded volume as the system. Suppose the system of fluid gets displaced from the position1 shown in figure to that in position 2 in a small time interval.
- Now, we shall find out the work done by different forces to use Work-Kinetic Energy theorem.
- Here four forces are acting on the system. Normal force from wall, force  $P_{A_{1}}^{A}$  on left portion, force  $P_{2}^{A}_{2}$  on right portion and force of gravity mg.



• Work done  $W_{N}$  by normal force from the walls:  $W_{N} = 0$  ...(i)

(because normal force is perpendicular to motion of fluid)

- Work done  $W_1$  by force  $P_1A_1$  at the left end:  $W_1 = \text{force} \times \text{displacement} = P_1A_1\Delta x_1 \qquad \dots (ii)$
- Work done  $W_2$  by force  $P_2A_2$  at the right end:  $W_2 = \text{force} \times \text{displacement} = -P_2A_2\Delta x_2 \qquad \dots (iii)$ (negative sign because force  $P_2A_2$  & displacement  $\Delta x_2$  are in opposite directions)
- The work done on the system W<sub>3</sub>, by the gravitational force mg:

$$W_g = \text{force} \times \text{displacement} = mg(h_1 - h_2) \qquad \dots \text{(iv)}$$

• ∴ the total work done on the system, by using Work- KE theorem, we have:

$$W_{\rm T} = \Delta KE$$
  

$$\Leftrightarrow W_{\rm N} + W_{\rm 1} + W_{\rm 2} + W_{\rm g} = \frac{1}{2} m v_{\rm 2}^{2} - \frac{1}{2} m v_{\rm 1}^{2}$$
  

$$\Leftrightarrow P_{\rm 1}A_{\rm 1}\Delta x_{\rm 1} - P_{\rm 2}A_{\rm 2}\Delta x_{\rm 2} + mg(h_{\rm 1} - h_{\rm 2}) = \frac{1}{2} m v_{\rm 2}^{2} - \frac{1}{2} m v_{\rm 1}^{2}$$
  
... (v)

pressure.

2, we get

we have

 $\Leftrightarrow \rho g h = \frac{1}{2} \rho v_2^2$ 

 $= \sqrt{2gh}$ 

Torricelli's Theorem (Speed of Efflux)

shown in figure. Such a hole is called an orifice.

the liquid at position 1 and 2 respectively.

From the equation of continuity, we get

 $A_1 v_1 = A_2 v_2$  or  $v_1 = \frac{A_2}{A} v_2$ 

Consider liquid of density  $\rho$  in a tank of large cross sectional area  $A_1$ . There is a **very small hole** of crosssectional area  $A_2$  at the bottom with liquid flowing out as

Let  $v_1$  and  $v_2$  be the speed and  $P_1$  and  $P_2$  be the speed of

The idea here is that both the tank and the narrow opening (orifice) are open to the atmosphere. The pressure will be the same at 1 and 2 because they are open to the atmosphere. Therefore,  $P_1 = P_2 = P_0 =$  atmospheric

As  $A_1 >> A_2$ , so the liquid may be taken at rest at the top, *i.e.*,  $v_1 = 0$ . Applying Bernoulli's equation at points 1 and

Now, applying Bernoulli's equation at positions 1 and 2,

(speed of efflux)

The speed of liquid coming out though a small hole (orifice) at a depth 'h' below the free surface is same as that of a particle fallen freely through the height 'h' under

The speed of the liquid coming out is called the speed of

gravity. This is known as Torricelli's theorem.

 $P_1 + \frac{1}{2}\rho v_1^2 + \rho g h_1 = P_2 + \frac{1}{2}\rho v_2^2 + \rho g h_2$ 

 $\Leftrightarrow P_0 + \frac{1}{2}\rho(0)^2 + \rho g(h) = P_0 + \frac{1}{2}\rho v_2^2 + \rho g(0)$ 

 $P_1 = P_0$ 

- ✓ The second term  $\frac{1}{2} \frac{V^2}{g}$  is called 'velocity head'.
- $\checkmark$  The third term h is known as the 'elevation head'.

efflux.

7488044834

 $P_2 = P_c$ 

#### Venturimeter

- **Venturimeter:** It is a device used to measure the rate of flow of a liquid through a pipe. It is an application of *Bernoulli's principle*. It is also called flow meter or venture tube.
- Construction. It consists of a horizontal tube having wider opening of cross-section A<sub>1</sub> and a narrow neck of cross-section A<sub>2</sub>. These two regions of the horizontal tube are connected to a manometer, containing a liquid of density ρ<sub>m</sub>.



• Working. Let the liquid velocities be  $v_1$  and  $v_2$  at the wider and the narrow portions. Let  $P_1$  and  $P_2$  be the liquid pressures at these regions. By the equation of continuity,

$$A_1 v_1 = A_2 v_2 \implies \frac{A_1}{A_2} = \frac{v_2}{v_1}$$

• If the liquid has density  $\rho$  and is flowing horizontally, then from Bernoulli's equation,

$$P_{1} + \frac{1}{2}\rho v_{1}^{2} + \rho g h_{1} = P_{2} + \frac{1}{2}\rho v_{2}^{2} + \rho g h_{2}$$
  
or  $P_{1} + \frac{1}{2}\rho v_{1}^{2} = P_{2} + \frac{1}{2}\rho v_{2}^{2}$  (::  $h_{1} = h_{2}$ )  
or  $P_{1} - P_{2} = \frac{1}{2}\rho \left(v_{2}^{2} - v_{1}^{2}\right)$   
 $= \frac{1}{2}\rho v_{1}^{2} \left(\frac{v_{2}^{2}}{v_{1}^{2}} - 1\right)$   
 $= \frac{1}{2}\rho v_{1}^{2} \left(\frac{A_{1}^{2}}{A_{2}^{2}} - 1\right)$  [::  $\frac{A_{1}}{A_{2}} = \frac{v_{2}}{v_{1}}$ ]  
 $= \frac{1}{2}\rho v_{1}^{2} \left(\frac{A_{1}^{2} - A_{2}^{2}}{A_{2}^{2}}\right)$ 

• If *h* is the height difference in the two arms of the manometer tube, then  $P - P = o_{1}hg$ 

• 
$$\therefore \rho_m hg = \frac{1}{2} \rho v_1^2 \left( \frac{A_1^2 - A_2^2}{A_2^2} \right)$$
  
 $\therefore v_1 = \sqrt{\frac{2h\rho_m g}{\rho} \times \frac{A_2^2}{A_1^2 - A_2^2}}$ 

• Volume flow rate of the liquid,

$$Q = A_1 v_1 = A_1 A_2 \sqrt{\frac{2h\rho_m g}{\rho (A_1^2 - A_2^2)}}$$

#### Viscosity

- Informally, viscosity is the quantity that describes a fluid's resistance to flow.
- Fluids resist the relative motion of immersed objects through them as well as to the motion of layers with differing velocities within them.
- In a laminar flow, the relative velocity between the consecutive layers of fluid surfaces results in tangential force at the of the layers known as viscous force and the property of the fluid causing it is known as viscosity.



- The layer of the liquid in contact with the surface remain stuck to it due to adhesive force and has zero velocity. The velocity of layer gradually increases on moving upwards from the surface and is the largest at the top.
- If a liquid flow easily, it means it has less viscosity; e.g., kerosene is less viscous than diesel. Similarly, honey is more viscous than water.
- Hard materials such as rock can be considered as liquids, because they can flow although extremely slowly. Glass windows in very old buildings are often thicker at the bottom than the top because over hundreds of years the glass has flowed downwards.

#### **Coefficient of viscosity**

• Consider the steady flow of liquid on some horizontal stationary surface as shown in the figure.



• According to Newton, the viscous force F between two adjacent layers of a laminar flow at a given temperature is

- i. directly proportional to the area (A) of the layers in contact  $F \propto A$  ...(i)
- ii. directly proportional to the velocity gradient  $\frac{dv}{dt}$

$$F \propto \frac{dv}{dx}$$
 ...(ii)

Combining (i) & (ii)

$$F \propto A \frac{dv}{dx}$$
  
 $\Leftrightarrow F = -\eta A \frac{dv}{dx}$  (viscous force)

where  $\eta$  is the constant of proportionality known as the coefficient of viscosity of the fluid. Its magnitude depends on the type of the fluid and its temperature.

- Negative sign states that the direction of viscous force is opposite to that of relative velocity of the layer wrt another layer.
- **S.I. Unit of**  $\eta$  : Ns/m<sup>2</sup> which is same as Pa-s
- CGS Unit of  $\eta$  : dyne-s/cm<sup>2</sup> which is also known as poise.

1 Pa-s = 10 poise

• Its dimensional formula is  $M L^{-1} T^{-1}$ .

#### Definition of $\boldsymbol{\eta}$

• Taking 
$$A = 1$$
 square unit and  $\frac{dv}{dx} = 1$  unit in the equation

$$F = \eta A \frac{dv}{dy}$$
, we get  $\eta = F$ 

Thus, the coefficient of viscosity can be defined as the viscous force acting per unit surface area of contact and per unit velocity gradient between two adjacent layers in a laminar flow of a fluid.

• Note that the co-efficient of viscosity of liquids decrease with increase in temperature, while that of gases increase with the increase in temperature.

#### Stokes Law

- When an object moves through a fluid, it experiences a viscous force which acts in opposite direction of its velocity.
- The resistive force (viscous force) on a small, smooth, solid spherical body of radius *R*, moving with velocity *v* through a laminar viscous medium of large dimensions, having co-efficient of viscosity η is given by

 $F = 6 \pi \eta R v \qquad \text{(Stokes law)}$ 

- This equation is called Stokes' Law which can be verified using dimensional analysis. This relationship is valid only for 'laminar flow'.
- This viscous force F acts opposite to velocity *v* of the object.

#### **Proof:**

• Viscous force depends on R, v and  $\eta$ .

1

• Let  $F = kR^a v^b \eta^c$ 

$$\Rightarrow [F] = k [R]^{a} [v]^{b} [\eta]^{c}$$
$$\Rightarrow [MLT^{-2}] = k [L]^{a} [LT^{-1}]^{b} [ML^{-1}T^{-1}]^{c}$$

• Comparing coefficients of *M*, *L* and *T* c = 1

$$a + b - c =$$

- -b-c = -2
- Solving we get,
- a = b = c = 1.• Thus,
  - $F = kR^1 v^1 \eta^1 = kR v \eta$
- Value of k is  $6\pi$ . Therefore,

 $F = 6\pi\eta Rv$ 

#### Terminal velocity

- In fluid dynamics, terminal velocity or settling velocity is the velocity at which the net force acting on an object moving through fluid becomes zero.
- Terminal velocity varies directly with the ratio of viscosity to weight. More viscosity means a lower terminal velocity, while increased weight means a higher terminal velocity.

#### Terminal velocity of a small sphere

- Suppose a small, smooth, solid sphere of radius r of material having density  $\rho$  falls freely in a laminar fluid of density  $\sigma$  ( <  $\rho$ ) and co-efficient of viscosity  $\eta$  as shown in the figure.
- Let its terminal velocity be v in the downward direction.
- The FBD in this figure lists three forces acting on the sphere:

.(ii)

• Weight *mg* downward: Therefore, weight

$$mg = \left(\frac{4}{3}\right)\rho\pi r^3g$$
 ...(i)

• Buoyant force *B* upward:

$$\Rightarrow B = \left(\frac{4}{3}\right)\pi r^3\sigma g \qquad \dots$$

 $B = V_{im} \rho_{liquid} g$ 

σ



#### **Surface Tension**

- Liquids sometimes form drops, and sometimes spread over a surface and wet it. Why does this happen, and why are raindrops never a metre wide?
- The reason is that **a fluids try to occupy minimum surface area**. This is because of a fluid property known as surface tension.

#### **Definition of surface tension**

- Let us consider an imaginary line *AB* drawn in any direction in a liquid surface. The surface on either side of the liquid exerts a pulling force *F* on the other side.
- This force F is perpendicular to the line AB and tangential to surface of the fluid.



- We give the definition of surface tension as: "The force exerted by the molecules lying on one side of an imaginary line of unit length, on the molecules lying on the other side of the line, which is perpendicular to the line and parallel to the surface is defined as the surface tension (S) of the liquid."
- In simple words, surface tension is perpendicular force (from either side of line *AB*) per unit length.
- Thus, if *F* be the force acting on either side of the line *AB* of length *L*, then the surface tension *S* is given by:

$$S = \frac{F}{L}$$
 (Surface tension)

Clearly, the SI unit of surface tension is N/m

#### Surface Potential Energy

- Surface energy. The free surface of a liquid possesses minimum area due to surface tension. To increase the surface area, molecules have to be brought from interior to the surface. Work has to be done against the forces of attraction. This work is stored as the potential energy of the molecules on the surface. So the molecules at the surface have extra energy compared to the molecules in the interior.
- The extra energy possessed by the molecules of surface film of unit area compared to the molecules in the interior is called surface energy. It is equal to the work done in increasing the area of the surface film by unit amount.

Surface energy = 
$$\frac{\text{work done}}{\text{increase in surface area}}$$

• Surface tension can also be defined as "the potential energy (U) stored in the surface of the liquid per unit area."

 $S = \frac{U}{U}$ 

⇔

- U = AS (surface potential energy)
- By this definition, its SI unit is Jm<sup>-2</sup> which is the same as Nm<sup>-1</sup>.

#### **Drops and bubbles**

- Let us assume that the pressures inside and outside are  $P_i$ and  $P_0$  ( $P_i > P_0$ ) respectively.
- The pressure on the concave surface is always more than that on the convex surface.
- Let the surface tension of the liquid forming the wall of the bubble be *T*.
- Suppose, on blowing the bubble, its radius increases from *R* to *R*+*dR*. The work done in this process can be calculated in two ways.
- 1 way:
  - $W = (force) \times (displacement)$
- $\Leftrightarrow$  W = pressure difference  $\times$  area  $\times$  displacement

$$\Leftrightarrow W = (P_i - P_i) 4 \pi R^{i} dR \dots (i)$$

- The surface area of the bubble of radius R is,  $S = 4 \pi R^2$
- $\therefore$  the increase in the surface area is,
- $dS = 4 \pi (R + dR)^2 4 \pi R^2$
- $\Leftrightarrow dS = 4 \pi (R^2 + dR^2 + 2R. dR R^2)$
- $\Leftrightarrow dS = 4 \pi (dR^2 + 2R. dR)$
- $\Leftrightarrow$  dS = 8  $\pi$  R. dR

(because dR is very small, we can ignore it )

 $P_{\rm o}$ 

#### **Fluid Mechanics**

•	$W = surface tension \times total increase in area$
\$ •	$W = 8 \pi S R dR$ (ii) Equating equations (i) and (ii), we get:
	$(P_{i} - P_{o}) 4 \pi R^{2} \cdot dR = 8 \pi S R dR$
€	$p_i - p_o = \frac{2S}{R}$ (Excess pressure inside a liquid drop)
•	For a soap bubble which has tow surface areas,
	$p_i - p_o = \frac{4S}{R}$ (Excess pressure inside a soap bubble)
•	Try to prove it yourself. Hint: Equation (ii) will be
	$W=16\piSRdR$ as soap bubble has two surfaces.
	Drops and bubbles (Ouick recap)

- The pressure on the concave surface is always more than that on the convex surface. P > P
- For the case of **air bubble inside water**, only one surface is formed. Therefore, for **air bubble** (or for any bubble or

drop where single surface is formed)  $p_i - p_o = \frac{2S}{R}$ 

• For the case of **soap bubble** where two surfaces are formed:  $n = n = \frac{4S}{2}$ 

formed:  $p_i - p_0 = \frac{4S}{R}$ 

#### Shape of Liquid Surface

- You must have seen water wets a glass container whereas mercury does not. Why?
- Before that let us familiarize with two new kind of intermolecular forces.:
- **Cohesive force :** Inter-molecular attractive force between molecules of the same matter.
- Adhesive force : Attractive force between molecules of different matters.
- Water molecules, for instance, are more attracted to glass than they are to one another. *It means in case of water cohesive forces are stronger than adhesive forces*. Water will therefore climb up a narrow glass tube that is dipped into a beaker of water, because the water would rather be in contact with the glass than with itself.
- Mercury molecules, on the other hand, are more attracted to each other than they are to glass. *It means in case of mercury cohesive forces are lesser than adhesive forces.* Mercury will avoid contact with a narrow glass tube that is dipped into a beaker of mercury.

#### **Capillary Action and Contact angle**

- **Capillarity:** Liquids display a behavior called **capillarity** or **capillary action** (capillary is a kind of narrow glass tube) because their molecules are more or less attracted to the surface they contact than they are to themselves.
- "The phenomenon of rise or fall of a liquid in a capillary, held vertical in a liquid, due to its property of surface tension is called capillarity."
- Capillary action is the result of surface tension and adhesive forces.



• **Contact Angle:** The tangent drawn at a point , where the surface of meniscus is in contact with wall of the capillary, makes an angle  $\theta$  with the wall.  $\theta$  is known as the contact angle of the liquid with the matter of the capillary.



- Case 1: The adhesive forces (liquid-glass) are greater than the cohesive forces (liquid-liquid)
- The liquid clings to the walls of the container the liquid "wets" the surface, e.g., water. The meniscus of water in the capillary is **concave**.
- In this case contact angle,  $\theta < 90^{\circ}$ .
- Case 2: The adhesive forces (liquid-glass) are lesser than the cohesive forces (liquid-liquid)
- The liquid curves downward the liquid does not "wet" the surface, e.g., mercury. The meniscus of mercury in the capillary is **convex.**
- In this case contact angle,  $\theta > 90^{\circ}$ .

#### **CAPILLARY RISE**

- We are going to derive capillary rise *h* (the height or depth) above which liquid rises or falls in a capillary tube when it is inserted in fluid).
- Suppose liquid rises to height *h* in a capillary of radius *r* held vertical in the liquid as shown in the figure.
- The radius of concave meniscus of liquid in the capillary is *R*.



• From second figure, it is clear that

$$\cos\theta = \frac{r}{R}$$

 $\Rightarrow R = \frac{r}{\cos \theta} \qquad \dots (i)$ 

• The pressure on the concave surface of the meniscus (P<sub>o</sub>) is greater than the pressure on the convex surface (P<sub>i</sub>).

$$\therefore P_o - p_i = \frac{2S}{R} \qquad \dots \text{ (ii)}$$

( because, the liquid has one free surface.)

• Also, for equilibrium, the pressure at point B is the same as at point A which is  $P_o$  as both are at the same horizontal level.

$$\therefore P_{a} - P_{i} = h \rho g \qquad \dots \text{(iii)}$$

where  $\rho$  = density of the liquid and g = acceleration due to gravity.

• Comparing equations (2) and (3), 28

$$\frac{2S}{R} = h\rho g$$

⇔

$$h = \frac{2RS}{\rho g} = \frac{2S\cos\theta}{r\rho g}$$

(putting the value of R from equation (i))

• Hence, capillary rise is given by

 $h = \frac{2S\cos\theta}{r\rho g}$  (Capillary rise)

• For mercury and glass,  $\theta > 90^\circ$ . Hence,  $\cos \theta$  is negative. Therefore, mercury falls in a glass capillary and its meniscus is convex.

#### **Mechanical Properties of Matter**

#### Elasticity

- **Elasticity:** If a body regains its original size and shape after the removal of deforming force, it is said to be elastic body and this property is called elasticity.
- **Perfectly elastic body:** If a body regains its original size and shape completely and immediately after the removal of deforming force, it is said to be a perfectly elastic body. The nearest approach to a perfectly elastic body is quartz fibre.
- **Plasticity:** If a body does not regain its original size and shape even after the removal of deforming force, it is said to be a plastic body and this property is called plasticity.
- **Perfectly plastic body:** If a body does not show any tendency to regain its original size and shape even after the removal of deforming force, it is said to be a perfectly plastic body. Putty and paraffin wax are nearly perfectly plastic bodies.
- Note: No body is perfectly elastic or perfectly plastic. All the bodies found in nature lie between these two limits. When the elastic behavior of a body decreases, its plastic behavior increases.

#### Stress

• **Stress:** *The internal restoring force set up per unit area of cross-section of the deformed body is called stress.* As the restoring force is equal and opposite to the external deforming force *F* under equilibrium, therefore

Stress = 
$$\frac{F}{A}$$
 (stress)

The SI unit of stress is  $Nm^{-2}$  and the CGS unit is dyne  $cm^{-2}$ . The dimensional formula of stress is  $[ML^{-1}T^{-2}]$ .

- Types of stress:
- (a) **Tensile stress:** It is the restoring force set up per unit cross-sectional area of a body when the length of the body increases in the direction of the deforming, force. It is also known as *longitudinal stress*.
- (b) **Compressive stress:** It is the restoring force set up per unit cross-sectional area of a body when its length decreases under a deforming force.
- (c) **Hydrostatic stress**: If a body is subjected to a uniform force from all sides, then the corresponding stress is called *hydrostatic stress* or *volume stress*.
- (d) **Tangential or Shearing stress:** When a deforming force acts tangentially to the surface of a body, it produces a change in the shape of the body. The tangential force applied per unit area is equal to the tangential stress.

#### **Fluid Mechanics**

Strain

**Strain:** The ratio of the change in any dimension produced in the body to the original dimension is called strain.

Strain –	Change in dimension
Origina	Original dimension

Strain has no units and dimensions.

Types of strain

(a) **Longitudinal strain:** It is defined as the increase in length per unit original length, when the body is deformed by external forces.

Longitudinal strain =	Change in length	$\Delta l$
	Original length	l

(b) **Volumetric strain**: It is defined as the change in volume per unit original volume, when the body is deformed by external forces.

Volumetric strain =  $\frac{\text{Change in volume}}{\text{Original volume}} = \frac{\Delta V}{V}$ 

(c) **Shear strain:** It is defined as the angle  $\theta$  (in radian), through which a face originally perpendicular to the fixed face gets turned on applying tangential deforming force.

Shear strain =  $\theta$  = tan  $\theta$ =  $\frac{\text{Relative displacement between 2 parallel planes}}{\text{Distance between parallel planes}}$ 

• **Elastic limit:** The maximum stress within which the body regains its original size and shape after the removal of deforming force is called elastic limit. If the deforming force exceeds the elastic limit, the body acquires a permanent set or deformation and is said to be *overstrained*.

#### HOOKE'S LAW & MODULUS OF ELASTICITY

- **Hooke's law:** It states that within the elastic limit, the stress is directly proportional to strain. Thus within the elastic limit,
  - Stress  $\propto$  Strain  $\Rightarrow \frac{\text{Stress}}{\text{Stress}} = \text{Constant}$

 $\Rightarrow \frac{1}{\text{Strain}} = \text{Constant}$ 

**Modulus of elasticity**: *The modulus of elasticity or coefficient of elasticity of a body is defined as the ratio of* stress to the corresponding strain, within the elastic limit.

 $E = \frac{\text{Stress}}{\text{Strain}}$ 

The SI unit of modulus of elasticity is  $Nm^{-2}$  and its dimensions are  $[ML^{-1}T^{-2}]$ .

R =

Units and dimensions of moduli of elasticity: The SI unit of moduli of elasticity is  $Nm^{-2}$  and its CGS unit is dyne cm<sup>-2</sup>. Its dimensional formula is  $[ML^{-1}T^{-2}]$ . Its value depends on the nature of the material of the body and the manner in which it is deformed.

- **Different types of moduli of elasticity**.: Corresponding to the three types of strain, we have three important moduli of elasticity:
- (a) Young's modulus (Y): Within the elastic limit, the ratio of longitudinal stress to the longitudinal strain is called Young's modulus. Thus,

$v_{\rm Longitudinal Stress}$	(Voung's modulus)
$T = \frac{1}{\text{Longitudinal Strain}}$	(Toung S modulus)

$$\therefore Y = \frac{F / A}{\Delta l / l}$$
$$\Rightarrow Y = \frac{F}{A} \cdot \frac{l}{\Delta l}$$

(b) Modulus of rigidity or shear modulus or torsional modulus ( $\eta$ ): Within the elastic limit, the ratio of shear stress to shear strain is called modulus of rigidity. Thus



- Shear strain  $= \theta \approx \tan \theta = \frac{AA'}{AB} = \frac{\Delta l}{l}$
- The modulus of rigidity is given by

$n_{\rm m}$ Shear stress	F/A	F	F l
$\sqrt{-\frac{1}{5}}$ Shear strain	$\theta$	$\overline{A\theta}$	$\overline{A} \cdot \overline{\Delta l}$

- (c) **Bulk modulus** (*B*): Within the elastic limit, the ratio of volume stress to the volumetric strain is called bulk modulus of elasticity.
- Consider a body of volume V and surface area A. Suppose a force F acts uniformly over the whole surface of the body and it decreases the volume by  $\Delta V$ , then bulk modulus of elasticity is given by

Volumetric stress Volumetric strain (Bulk modulus)

#### **Fluid Mechanics**

# $\therefore B = -\frac{F/A}{\Delta V/V} = \frac{F}{A} \cdot \frac{V}{\Delta V}$ $\Rightarrow B = -\frac{PV}{\Delta V}$

where p (= F/A) is the normal pressure. Negative sign shows that the volume decreases with the increase in stress.

• **Compressibility** (*K*): The reciprocal of the bulk modulus of a material is called its compressibility. Thus,

 $K = \frac{1}{B} = -\frac{\Delta V}{pV}$  (Compressibility)

#### **Poisson's ratio**

- When a wire is loaded, its length increases but its diameter decreases. The strain produced in the direction of applied force is called longitudinal strain and that produced in the perpendicular direction is called lateral strain.
- **Definition:** Within the elastic limit, the ratio of lateral strain to the longitudinal strain is called Poisson's ratio. Suppose the length of the loaded wire increases from *l* to  $l+\Delta l$  and its diameter decreases from *D* to  $D-\Delta D$ .



- The negative sign indicates that longitudinal and lateral strains are in opposite sense.
- As the Poisson's ratio is the ratio of two strains, it has no units and dimensions.

#### SPACE FOR NOTES



- Figure shows a stress-strain curve for a metal wire which is gradually being loaded.
- (a) The initial part *OA* of the graph is a straight line indicating that stress is proportional to strain. Upto the point *A* Hooke's law is obeyed. The point *A* is called the **proportional limit.** In this region, the wire is perfectly elastic.
- (b) After the point *A*, the stress is not proportional to strain and a curved portion *AB* is obtained. However, if the load is removed at any point between *O* and *B*, the curve is retraced along *BAO* and the wire attains its original length. The portion *OB* of the graph is called **elastic region** and the point *B* is called **elastic limit or yield point**. The stress corresponding to the yield point is called yield strength ( $S_y$ ).



- (c) Beyond the point *B*, the strain increases more rapidly than stress. If the load is removed at any point *C*, the wire does not come back to its original length but traces dashed line *CE*. Even on reducing the stress to zero, a residual strain equal to *OE* is left in the wire. The material is said to have acquired a **permanent set**. The fact that the stress-strain curve is not retraced on reversing the strain is called **elastic hysteresis**.
- (d) If the load is increases beyond the point *C*, there is large increase in the strain or the length of the wire. In this region, the constrictions (called necks and waists) develop at few points along the length of the wire and the wire ultimately breaks at the point *D*, called the **fracture point.** In the region between *B* and *D*, the length of wire goes on increasing even without any addition of load. This region is called *plastic region* and the material is said to undergo *plastic flow* or *plastic deformation*. The stress corresponding to the breaking point is called *ultimate strength* or *tensile strength* of the material.

Elastic potential energy of a stretched wire				
To prove:				
Elastic potential energy $=\frac{1}{2} \times \text{Stress} \times \text{strain} \times \text{volume}$				
Proof:				
Suppose a force <i>F</i> applied on a wire of length <i>l</i> increases its length by $\Delta l$ . Initially, the internal restoring force in the wire is zero. When the length is increased by $\Delta l$ , the internal force increases from zero to <i>F</i> (= applied force). $\therefore$ Average internal force for an increase in length $\Delta l$ of wire $= \frac{0+F}{2} = \frac{F}{2}$ Work done on the wire is				
W = Average force × increase in length = $\frac{F}{2} \times \Delta l$				
• This work done is stored as elastic potential energy U in the wire.				
$\therefore U = \frac{1}{2}F \times \Delta l = \frac{1}{2}$ Stretching force × increase in length				
Let A be the area of cross-section of the wire. Then $\therefore U = \frac{1}{2} \frac{F}{A} \times \frac{\Delta l}{l} \times Al$				
$\therefore U = \frac{1}{2} \times \text{Stress} \times \text{strain} \times \text{volume}$				
Elastic potential energy per unit volume of the wire or				
elastic energy density is				
$u = \frac{U}{\text{Volume}}$				
or $u = \frac{1}{2}$ stress × strain				
But stress = Young's modulus $\times$ strain				
$\therefore u = \frac{1}{2} \times \text{Young's modulus} \times \text{strain}^2$				

#### SPACE FOR NOTES





### Physics Classes by Pranjal Sir

#### (Admission Notice for XI & XII - 2014-15)

#### **Batches for Std XIIth**

- Batch 1 (Board + JEE Main + Advanced): (Rs. 16000)
- Batch 2 (Board + JEE Main): (Rs. 13000)
- Batch 3 (Board): (Rs. 10000)
- Batch 4 (Doubt Clearing batch): Rs. 8000



Bokaro Centre

Address:JB 20, Near Jitendra Cinema, City Centre, Sec 4, Bokaro Mobile: 7488044834 Website: www.vidyadrishti.org

#### About P. K. Bharti Sir (Pranjal Sir)

- B. Tech., IIT Kharagpur (2009 Batch)
- H.O.D. Physics, Concept Bokaro Centre
- Visiting faculty at D. P. S. Bokaro
- Produced AIR 113, AIR 475, AIR 1013 in JEE Advanced
- Produced AIR 07 in AIEEE (JEE Main)

Address: Concept, JB 20, Near Jitendra Cinema, Sec 4, Bokaro Steel City Ph: 9798007577, 7488044834 Email: <u>pkbharti.iit@gmail.com</u> Website: www.vidyadrishti.org

#### Physics Class Schedule for Std XIIth (Session 2014-15) by Pranjal Sir

Sl. No.	Main Chapter	Topics	Board level	JEE Main Level	JEE Adv Level
Basics from XIth Vectors, FBD, Work, Energy, Rotation, SHM		3 <sup>rd</sup> Mar to 4 <sup>th</sup> Apr 14			
1.	Electric Charges and	Coulomb's Law	5 <sup>th</sup> & 6 <sup>th</sup> Apr	5 <sup>th</sup> & 6 <sup>th</sup> Apr	5 <sup>th</sup> & 6 <sup>th</sup> Apr
	Fields	Electric Field	10 <sup>th</sup> & 12 <sup>th</sup> Apr	10 <sup>th</sup> & 12 <sup>th</sup> Apr	10 <sup>th</sup> & 12 <sup>th</sup> Apr
		Gauss's Law	13 <sup>th</sup> & 15 <sup>th</sup> Apr	13 <sup>th</sup> & 15 <sup>th</sup> Apr	13 <sup>th</sup> & 15 <sup>th</sup> Apr
		Competition Level	NA	17 <sup>th</sup> & 19 <sup>th</sup> Apr	17 <sup>th</sup> & 19 <sup>th</sup> Apr
2.	<b>Electrostatic Potential</b>	Electric Potential	20 <sup>th</sup> & 22 <sup>nd</sup> Apr	20 <sup>th</sup> & 22 <sup>nd</sup> Apr	20 <sup>th</sup> & 22 <sup>nd</sup> Apr
	and Capacitance	Capacitors	24 <sup>th</sup> & 26 <sup>th</sup> Apr	24 <sup>th</sup> & 26 <sup>th</sup> Apr	24 <sup>th</sup> & 26 <sup>th</sup> Apr
		Competition Level	NA	27 <sup>th</sup> & 29 <sup>th</sup> Apr	27 <sup>th</sup> & 29 <sup>th</sup> Apr, 1 <sup>st</sup> , 3 <sup>rd</sup> & 4 <sup>th</sup> May
PART	TEST 1	Unit 1 & 2	4 <sup>th</sup> May	NA	NA
			NA	11 <sup>th</sup> May	11 <sup>th</sup> May
3.	Current Electricity	Basic Concepts, Drift speed, Ohm's Law, Cells, Kirchhoff's Laws, Wheatstone bridge, Ammeter, Voltmeter, Meter Bridge, Potentiometer etc.	6 <sup>th</sup> , 8 <sup>th</sup> , 10 <sup>th</sup> , 13 <sup>th</sup> May	6 <sup>th</sup> , 8 <sup>th</sup> , 10 <sup>th</sup> , 13 <sup>th</sup> May	6 <sup>th</sup> , 8 <sup>th</sup> , 10 <sup>th</sup> , 13 <sup>th</sup> May
		Competition Level	NA	15 <sup>th</sup> & 16 <sup>th</sup> May	15 <sup>th</sup> , 16 <sup>th</sup> , 17 <sup>th</sup> , 18 <sup>th</sup> & 19 <sup>th</sup> May
PART	TEST 2	Unit 3	18 <sup>th</sup> May	NA	NA
			NA	20 <sup>th</sup> May	20 <sup>th</sup> May
SUMM	ER BREAK	21 <sup>st</sup> May 2013 to	30 <sup>th</sup> May 2013		
4.	Moving charges and Magnetism	Force on a charged particle (Lorentz force), Force on a current carrying wire, Cyclotron, Torque on a current carrying loop in magnetic field, magnetic moment	31 <sup>st</sup> May, 1 <sup>st</sup> & 3 <sup>rd</sup> Jun	31 <sup>st</sup> May, 1 <sup>st</sup> & 3 <sup>rd</sup> Jun	31 <sup>st</sup> May, 1 <sup>st</sup> & 3 <sup>rd</sup> Jun
		Biot Savart Law, Magnetic field due to a circular wire, Ampere circuital law, Solenoid, Toroid Competition Level	5 <sup>th</sup> , 7 <sup>th</sup> & 8 <sup>th</sup> Jun NA	5 <sup>th</sup> , 7 <sup>th</sup> & 8 <sup>th</sup> Jun 10 <sup>th</sup> & 12 <sup>th</sup> Jun	5 <sup>th</sup> , 7 <sup>th</sup> & 8 <sup>th</sup> Jun 10 <sup>th</sup> , 12 <sup>th</sup> , 14 <sup>th</sup> & 15 <sup>th</sup>
			, _th _		Jun
PART	TEST 3	Unit 4	15 <sup>th</sup> Jun	NA	NA

Fluid N	Techanics	Author. I. K. Dharti (D. Teen., II	T Kilaragpur), 11.0.	D. Thysics at Concep	t Bokaro Centre
			NA	22 <sup>nd</sup> Jun	22 <sup>nd</sup> Jun
5.	Magnetism and		17 <sup>th</sup> , 19 <sup>th</sup> & 21 <sup>st</sup>	17 <sup>th</sup> , 19 <sup>th</sup> & 21 <sup>st</sup>	Not in JEE Advanced
	Matter		Jun	Jun	Svllabus
6	Flootromognotio	Faraday's Laws, Lenz's Laws, A.C.	24 <sup>th</sup> 26 <sup>th</sup> & 28 <sup>th</sup>	24 <sup>th</sup> 26 <sup>th</sup> & 28 <sup>th</sup>	24 <sup>th</sup> 26 <sup>th</sup> & 28 <sup>th</sup> Jun
0.	Liecti omagnetic	Generator, Motional Emf. Induced	24,20 & 20 I	24,20 & 20 I	24,20 & 28 Juli
	Induction	Emf, Eddy Currents, Self Induction,	Jun	Jun	
		Mutual Induction			
		Competition Level	NA	29 <sup>th</sup> Jun & 1 <sup>st</sup>	29 <sup>th</sup> Jun, 1 <sup>st</sup> , 3 <sup>rd</sup> & 5 <sup>th</sup>
				Jul	Jul
PART 7	TEST 4	Unit 5 & 6	6 <sup>th</sup> Jul	NA	NA
			NA	13 <sup>th</sup> Jul	13 <sup>th</sup> Jul
7.	Alternating current	AC. AC circuit. Phasor. transformer.	8 <sup>th</sup> 10 <sup>th</sup> & 12 <sup>th</sup>	8 <sup>th</sup> 10 <sup>th</sup> & 12 <sup>th</sup>	g <sup>th</sup> 10 <sup>th</sup> & 12 <sup>th</sup> Inl
	After hatting current	resonance,	0,10 & 12	0,10 & 12	8,10 &12 Jui
		Compatition Laval			
0		Competition Level		15 July	
ð.	Electromagnetic		19 <sup>44</sup> & 20 <sup>44</sup> July	19 <sup></sup> & 20 <sup></sup> July	Not in JEE Advanced
	Waves		th	th	Syllabus
PART 1	TEST 5	Unit 7 & 8	27 <sup>th</sup> Jul	27 <sup>th</sup> Jul	27 <sup>th</sup> Jul
Revision	n Week	Upto unit 8	$31^{\text{st}}$ Jul & $2^{\text{nd}}$	31 <sup>st</sup> Jul & 2 <sup>nd</sup>	31 <sup>st</sup> Jul & 2 <sup>nd</sup> Aug
			Aug	Aug	
Grand T	Test 1	Upto Unit 8	3 <sup>rd</sup> Aug	3 <sup>rd</sup> Aug	3 <sup>rd</sup> Aug
0		Reflection	5th e 7th Ang	5th e 7th Ana	5 <sup>th</sup> 8 7 <sup>th</sup> And
,		Defraction	oth e 12th Arres	oth e 10th Aug	$5 \propto 7 \text{Aug}$
	Der Orther	Refraction	9 & 12 Aug	9 & 12 Aug	9 & 12 Aug
	Ray Optics	Prism	14 <sup>th</sup> Aug	14 <sup>th</sup> Aug	14 <sup>th</sup> Aug
		Optical Instruments	16 <sup>th</sup> Aug	16 <sup>th</sup> Aug	Not in JEE Adv
					Syllabus
		Competition Level	NA	19 <sup>th</sup> & 21 <sup>st</sup> Aug	$19^{\text{th}}, 21^{\text{st}}, 23^{\text{rd}}, 24^{\text{th}}$
					Aug
10.		Huygens Principle	26 <sup>th</sup> Aug	26 <sup>th</sup> Aug	26 <sup>th</sup> Aug
		Interference	28 <sup>th</sup> & 30 <sup>th</sup> Aug	28 <sup>th</sup> & 30 <sup>th</sup> Aug	28 <sup>th</sup> & 30 <sup>th</sup> Aug
	Wave Optics	Diffraction	31 <sup>st</sup> Aug	31 <sup>st</sup> Aug	31 <sup>st</sup> Aug
	ľ	Polarization	2 <sup>nd</sup> Sen	2 <sup>nd</sup> Sen	2 <sup>nd</sup> Sen
		Competition Level	NA NA	1 <sup>th</sup> & 6 <sup>th</sup> Sen	$\frac{2}{4^{\text{th}}} \frac{50 \text{p}}{6^{\text{th}}} \frac{7^{\text{th}}}{7^{\text{th}}} \frac{9^{\text{th}}}{9^{\text{th}}} \frac{11^{\text{th}}}{11^{\text{th}}}$
				- au sep	+, 0, 7, 7, 11
	DADT TEST 6	Unit 9 & 10	1 Ath Com	14 <sup>th</sup> Com	J4 <sup>th</sup> Com
	FART TEST 0			14 Sep	14 Sep
0 17	REVISIO	N ROUND I (For JEE Main & JE	Le Advanced Level	$13^{-1}$ Sep to $27^{-1}$ Se	
Grand T	lest 2	Opto Unit 10	28 <sup></sup> Sep	28 <sup></sup> Sep	28 <sup></sup> Sep
DUSSE					
DUSSEHRA & d-ul-Zuha Holidays: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct					
20002	HRA & d-ul-Zuha Holida	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct			4
11.	HRA & d-ul-Zuha Holida Dual Nature of	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc	9 <sup>th</sup> & 11 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct
11.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc	9 <sup>th</sup> & 11 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct
11. Grand T	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10	<b>9<sup>th</sup> &amp; 11<sup>th</sup> Oct</b>	<b>9<sup>th</sup> &amp; 11<sup>th</sup> Oct</b>	<b>9<sup>th</sup> &amp; 11<sup>th</sup> Oct</b>
11. Grand T	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10	9 <sup>th</sup> & $11^{th}$ Oct $12^{th}$ Oct	<b>9<sup>th</sup> &amp; 11<sup>th</sup> Oct</b> 12 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct
11. Grand T 12.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Fest 3 Atoms	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct
11. Grand T 12.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct
11. Grand T 12.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct NA	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> & 25 <sup>th</sup> Oct
11. Grand T 12. 13.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct NA 26 <sup>th</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> & 25 <sup>th</sup> Oct NA
11. Grand T 12. 13.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Pagin Concents and Diodes	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct NA 26 <sup>th</sup> Oct 20 <sup>th</sup> 20 <sup>th</sup> 20 <sup>th</sup>	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA $24^{th}$ 20 <sup>th</sup> 20 <sup>th</sup>	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> & 25 <sup>th</sup> Oct NA
11. Grand T 12. 13.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct NA 26 <sup>th</sup> Oct 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct 8 <sup>th</sup> V	$9^{th}$ & 11^{th} Oct 12^{th} Oct 14^{th} & 16^{th} Oct 18^{th} & 19^{th} Oct 21^{st} Oct NA 26^{th}, 28^{th}, 30^{th}	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv
11. Grand T 12. 13.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct NA 26 <sup>th</sup> Oct 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           21 <sup>st</sup> & 25 <sup>th</sup> Oct           NA           Not in JEE Adv           Syllabus
11. Grand T 12. 13. 14.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           21 <sup>st</sup> & 25 <sup>th</sup> Oct           NA           Not in JEE           Not in JEE           Not in JEE
11. Grand T 12. 13. 14. 15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct NA 26 <sup>th</sup> Oct 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           21 <sup>st</sup> & 25 <sup>th</sup> Oct           NA           Not in JEE           Not in JEE           Adv           Syllabus           Not in JEE
11. Grand T 12. 13. 14. 15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov           9 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           21 <sup>st</sup> & 25 <sup>th</sup> Oct           NA           Not in JEE           Not in JEE           Adv           Syllabus           NA
11.       Grand T       12.       13.       14.       15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov 9 <sup>th</sup> Nov 8 <sup>th</sup> , 9 <sup>th</sup> & 11 <sup>th</sup>	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv         Syllabus         Not in JEE Adv         Syllabus         NA         8th, 9th, 11th, 13th &
11. Grand T 12. 13. 14. 15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov 9 <sup>th</sup> Nov 8 <sup>th</sup> , 9 <sup>th</sup> & 11 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           21 <sup>st</sup> & 25 <sup>th</sup> Oct           NA           Not in JEE Adv           Syllabus           Not in JEE Adv           Syllabus           NA           Soft in JEE Adv           Syllabus           NA           Not in JEE Adv           Syllabus           NA           Sth           Sth           9 <sup>th</sup> , 11 <sup>th</sup> , 13 <sup>th</sup> & 15 <sup>th</sup> Nov
11. Grand T 12. 13. 14. 15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13 PART TEST 9	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level Unit 11, 12, 13, X-Rays	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA           NA	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov 9 <sup>th</sup> Nov 8 <sup>th</sup> , 9 <sup>th</sup> & 11 <sup>th</sup> Nov 16 <sup>th</sup> Nov	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv         Syllabus         NA         Not in JEE Adv         Syllabus         NA         Soft in JEE Adv         Syllabus         NA         16th Nov
11.           Grand T           12.           13.           14.           15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13 PART TEST 9 Revision Round 2	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level Unit 11, 12, 13, X-Rays Mind Maps & Back up classes for late	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA           18 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov 9 <sup>th</sup> Nov 8 <sup>th</sup> , 9 <sup>th</sup> & 11 <sup>th</sup> Nov 16 <sup>th</sup> Nov 18 <sup>th</sup> Nov	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv         Syllabus         NA         8th, 9th, 11th, 13th & 15th Nov         16th Nov         18th Nov to Board
11. Grand T 12. 13. 14. 15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13 PART TEST 9 Revision Round 2 (Board Level)	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level Unit 11, 12, 13, X-Rays Mind Maps & Back up classes for late registered students	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA           NA           9 <sup>th</sup> Nov           NA           18 <sup>th</sup> Nov to           Board Exams	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov 9 <sup>th</sup> Nov 8 <sup>th</sup> , 9 <sup>th</sup> & 11 <sup>th</sup> Nov 16 <sup>th</sup> Nov 18 <sup>th</sup> Nov to Board Exams	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv         Syllabus         Not in JEE Adv         Syllabus         NA         8th, 9th, 11th, 13th &         15th Nov         16th Nov         18th Nov to Board Exams
11.       Grand T       12.       13.       14.       15.	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13 PART TEST 9 Revision Round 2 (Board Level) Revision Round 3	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level Unit 11, 12, 13, X-Rays Mind Maps & Back up classes for late registered students	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA           NA           18 <sup>th</sup> Nov           NA           18 <sup>th</sup> Nov           18 <sup>th</sup> Nov           18 <sup>th</sup> Nov           18 <sup>th</sup> Nov           18 <sup>th</sup> Nov	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov 9 <sup>th</sup> Nov 8 <sup>th</sup> , 9 <sup>th</sup> & 11 <sup>th</sup> Nov 16 <sup>th</sup> Nov 18 <sup>th</sup> Nov to Board Exams 18 <sup>th</sup> Nov to IEE	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv         Syllabus         NA         8th , 9th , 11th , 13th &         16th Nov         18th Nov to Board         Exams         18th Nov to IEE
11. Grand T 12. 13. 14. 15. (X	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13 PART TEST 9 Revision Round 2 (Board Level) Revision Round 3 (Ith portion for LEE)	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level Unit 11, 12, 13, X-Rays Mind Maps & Back up classes for late registered students	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA           NA           18 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA           18 <sup>th</sup> Nov to           Board Exams           18 <sup>th</sup> Nov to JEE	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           21 <sup>st</sup> Oct           NA           26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov           9 <sup>th</sup> Nov           8 <sup>th</sup> , 9 <sup>th</sup> & 11 <sup>th</sup> Nov           16 <sup>th</sup> Nov           18 <sup>th</sup> Nov to           18 <sup>th</sup> Nov to	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv         Syllabus         NA         8th, 9th, 11th, 13th &         15th Nov         16th Nov         18th Nov to Board Exams         18th Nov to JEE
11. Grand T 12. 13. 14. 15. 15. 15. 16. 17. 17. 17. 17. 17. 17. 17. 17. 17. 17	HRA & d-ul-Zuha Holida Dual Nature of Radiation and Matter Test 3 Atoms Nuclei X-Rays PART TEST 7 Semiconductors Communication System PART TEST 8 Unit 11, 12 & 13 PART TEST 9 Revision Round 2 (Board Level) Revision Round 3 Ith portion for JEE) 30 Full Test Series	ys: 29 <sup>th</sup> Sep to 8 <sup>th</sup> Oct Photoelectric effect etc Upto Unit 10 Unit 11, 12 & 13 Basic Concepts and Diodes, transistors, logic gates Unit 14 & 15 Competition Level Unit 11, 12, 13, X-Rays Mind Maps & Back up classes for late registered students Complete Syllabus	9 <sup>th</sup> & 11 <sup>th</sup> Oct           12 <sup>th</sup> Oct           14 <sup>th</sup> & 16 <sup>th</sup> Oct           18 <sup>th</sup> & 19 <sup>th</sup> Oct           NA           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Oct           26 <sup>th</sup> Nov           2 <sup>nd</sup> & 4 <sup>th</sup> Nov           9 <sup>th</sup> Nov           NA           18 <sup>th</sup> Nov to JEE           Date will be public	9 <sup>th</sup> & 11 <sup>th</sup> Oct 12 <sup>th</sup> Oct 14 <sup>th</sup> & 16 <sup>th</sup> Oct 18 <sup>th</sup> & 19 <sup>th</sup> Oct 21 <sup>st</sup> Oct NA 26 <sup>th</sup> , 28 <sup>th</sup> , 30 <sup>th</sup> Oct & 1 <sup>st</sup> Nov 2 <sup>nd</sup> & 4 <sup>th</sup> Nov 9 <sup>th</sup> Nov 9 <sup>th</sup> Nov 16 <sup>th</sup> Nov 18 <sup>th</sup> Nov to Board Exams 18 <sup>th</sup> Nov to JEE hed after Oct 2014	9th & 11th Oct         12th Oct         14th & 16th Oct         18th & 19th Oct         21st & 25th Oct         NA         Not in JEE Adv         Syllabus         NA         8th , 9th , 11th , 13th & 15th Nov         16th Nov         18th Nov to Board Exams         18th Nov to JEE